MODEL TEST PAPER

ENTRANCE EXAMINATION FOR ADMISSION TO B.Sc. (AGRICULTURE)

General Instructions for Students

- Every candidate should carry his/her valid Roll No. cum Admit Card to the Entrance Test. No candidate without the valid Roll No. cum Admit Card will be allowed to enter the examination centre.
- The question paper will be of Two Hours duration and will comprise of Hundred Multiple Choice Questions of One mark each.
- There will be four sections, viz; Physics, Chemistry. Biology OR Mathematics and General Awareness
 of the Subject.
- The candidates with 10+2 (Medical) will opt the section of Biology while the candidates with 10+2 (Non-Medical) will opt the Mathematics Section.
- The candidate has to mark the right option against the question number in the OMR sheet with black pen. The circles marked with pencil or blue pen will not be marked.
- 6. There will be no negative marking.
- 7. The OMR must be handed over to the Room Supervisor even if candidate has not filled any option.
- 8. No candidate will be allowed to leave the examination hall before two hours.
- 9. Don't write/make any identification marks(s)/religious symbols/slogan(s) on the answer books.
- The candidate must ensure that his OMR has been duly stamped.
- 11. Please ensure that you have signed the attendance sheet.
- Mobile Phones and other electronic gadgets such as Bluetooth etc. are strictly prohibited in the Examination Centre.

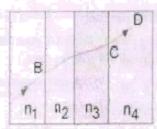
PHYSICS

1.	Hasmall amount of an	timony is added to	genna	mium	erystal:			
	 A) It becomes a p-type 	e semiconductor						
	B) The antirrony beed	ones an acceptor a	tom					
	C) There will be more	free electrons than	holes	in the	semiconducto	DE .		
	D) Its resistance is inc	reased						
2	he forward biasing of th	ne p-n junction						
	A) The positive termination	nal of the battery is	conn	ected t	op side and th	he depletio	n region becomes th	ńck
	B) The positive termin	ad of the battery is	come	ected t	on-side and th	ie depletion	region becomes th	in.
	C) The positive termin	nal of the battery is	conne	ested t	on-side and th	he depletion	a region becomes th	ick
	O) The positive termin	al of the battery is	conne	eted t	op-side and th	ne depletion	regior becomes th	in
j.	The electromagnetic ra	diation is caused by	y:					
	A) Stationary charge			B) U	niformly movi	ng charges		
	C) Accelerated charge	5		D) A	It of these			1
14.	The oscillating electric	and magnetic field	vecto	rs of c	lectromagnetic	e wave are	oriented along:	
	A) The same direction	and in phase						
	B) The same direction	but have a phase of	liffere	nce of	90°			
	C) Mutually perpendic	ular directions and	are in	phase	6			
	Dr Mutually perpendie	ular directions but	has a	phase	difference of	90°		
5.	The half life of radium after:	is about 1600 yea	rs, lf	100 g	of radium exis	sts new, 25	g will remain uncha	inged
	A) 3200 years	B) 4800 years		()	6400 years		D) 2400 years	
6.	The nuclii 7N ⁴⁴ and 6C	¹³ can be describe	das:					
	A) Isotones	B) Isobars		(C)	Isotopes		D) Isomers	
Ť.	A 226 volts input is su of the efficiency of the is:							
	A) 3.6 A	B) 2.8 A		C)	2.5 A		D) 5.0 A	
2.	One conducting U tube the tubes. The magnetithe figure. If each tube is then the omf induced his he width of each tu	ic field B is perpen moves towards the I in the circuit in te	ndicul other a	kar to t at a con	he plane of estant speed	À.		ween
1-	A) - B/v		R)	R/y			A. A.	I.
	CY 2B/v		D)	Zero		()		
	F-15							

9,	The flux linked w	ith a given coil at a	ny instant	t' is gi	ven by $p = 1$	0t ² – 50t	+ 250). Th	e inc	reed	emf a	ŧt
	A)-190 V	B) ~10 V		C)	10. V		D)	190	V			
10.	A bar magnet, of on it is	magnetic moment	M, is place	ed in m	nagnetic field	of indu	etion	в. Т	he to	asque	exerk	cel
	A) N · B	B) -Nt B		C)	$\overrightarrow{M} \times \overrightarrow{B}$		D)	8	× M			
n.	apart. A stationary	magnets are fixed y charge Q is place e O as shown in f	d at P in b	elweer	the gas of t	he .			90			
	 A) directed perpo 	endicular to the pla	ne of paper	rL T								
	Ез гею					S		М	0	No 1	8	
	C) directed along	OP							4	-		F
	B) directed along	P0					(-		CI			
12.	a long straight co	rrying a steady curr nductor corrying a vn in figure. The loc	steasy curr	ent l	at a distance				d		1	
	A net repulsive force away from the conductor											
	B) A net torque acting upward perpendicular to the horizontal plane											F
	C) A net torque a	eting downward no	ormal to the	horiz	ontal plane							
	Di A nel attractiv	ve force towards the	е совдцего	F						Æ:	٠.,	
13,		radius R, earlying I magnetic flux thro				th its cer	rtre					
	A) directly propo	ectional to 1	By	direc	tly proportion	nal to R						
	C) javersely prop	portional to R	D)	0.32								
14.	A current of 2A p	nassing through a co	onductor p	roduce	s 801 of heat	in 10 s	Ther	ęsist	апсс	of co	nduct	or or
	A) 0.5	B) 2		C)	4		D)	20				
15.	la electrolysis, the	e arrount of mass d	eposited or	libera	ued at an clas	etrode is	direct	y pr	opor	tional	ta:	
	A) Square of elec	ctrie charge	B)	Ато	unt of charge							
	(') Square of cur	ren l	D)	Cone	centration of	electroly	ke					
16.	In the circuit show	on in the figure the	potential di	Heren	ce between X	and Yv	vill be					
	At Zero Velis						F	10		X	7	1
	B) 20V							7				1
	C) 60V							120	¥		NO PE	1
	D) 120V							-[1]	-		VV	1
	The same							- 1	1		W 274	12.01

- Five resistances each of 5 Ω, are connected as shown in figure. If all the resistances are of 5 Ω the
 capitalent resistance between points (1) A & B and (2) A & C.
 - Α) (1) 7.5Ω (2) 2,25Ω
 - B) (1) 5\Omega (2) 25\Omega
 - C) (1) 2.5 \Omega (2) 3.1 \Omega
 - D) (I) 3Ω (2) 2.5Ω
- 18. The force between two point charges placed in vacuum is 18N at a separation of Imm. If a glass plate of thickness Imm and dielectric constant 6 be kept between the charges, then the force between them would be:
 - A) 18 N
- B) 108N
- C) 3 N
- D) 3×10-6N
- 19. The electric field inside a spherical shell of uniform surface charge density is:
 - A) Zero

- B) Constant, but non zero
- C) Proportional to the distance from centre
- D). None of the above
- 20. A point object is 24 cm above the surface of water (µ=4/3) in lake. A fish inside the water will observe the image to be at a point:
 - A) 6 cm above the surface of water
- B) 6 cm below the surface of water
- C) 18 cm above the surface of water
- D) 32 cm above the surface of water
- 21. A say of light passes through four transparent media with refractive indices n₁, n₂, n₃ and n₄ as shown in the figure. The surfaces of all media are parallel. If the emergent ray CD is parallel to the ireident ray AB, we must have:



- A) p1 102
- B) no=ny
- C) n₃=n₄
- D) $n_f = n_t$
- 22. The magnification m, the image position v and focal length f are related to one another by the relation :

A)
$$m = \frac{f - y}{f}$$

- B) $m = \frac{f}{f}$
- C) $m = \frac{f + v}{t}$
- D) $M = \frac{f}{v f}$

- 23. Which of the following statement is correct?
 - A) Photo-current increases with intensity of light
 - B) Photo-current is proportional to the applied voltage
 - C) Current in photocell increases with increasing frequency
 - D) Stopping potential increases with increase of incident light

4. An electron, proton and a car all have so	ame way	elength. The one possessin	ng high	est velocity is
A) Electron	B)	Proton		
C) Car	D)	All have same velocity		
5. In photocell, energy conversion is from				
A) Chemical to electrical	B)	Mechanical to electrical		
C) Optical to electrical	D)	Magnetic to electrical		
	CHEA	HSTRY		
The coordination number of a metal tha			bic pac	ked then stra
A) 6 . B) 12	1	C) 8	Di	
. The solution containing 18 g of glucose	per 100		100	
A) (), I molat B) ().1 molar		C) 1.8 molal		1.8 molar
When one Faraday is divided by Avoga	dro num			
A) Charge of electron in Coulombs		Charge of electron in esu		
(1) Charge on the nucleus		Current in Amperes.		
The units of rate constant for a zero ord		Control of the Contro		
A) mol L-1 sec 1 B) L mol 1 sec 1		C) ·see-1	D)	mol 11.
Which of the following will have highest				
A) Al ³⁺ icn B) PC ₄ ³⁻ ion	100	C) SO ₄ 2 ion	D)	Na ⁺ ion
Poling process is used for the removal of				
A) Al ₂ O ₃ from Al B) Cu ₂ O from t		C) Fe ₂ O ₁ from Fe	D)	Alf of these
Which of the following nitrogen oxide is		ly most stable :		
A) N ₂ O ₅ B) NO ₂		C) NO	10)	NaO
. Hybridization of S in SF ⁴ is:				
A) sp ² B) sp ²		C) sp ³ d	D)	sp ³ d ²
Which of the following halogers is oxidi	ized by n	itrie neid?		
A) I B) Br		C) CI	D)	Fair Inch
Bleaching powder is prepared from the	reaction	of:		
A) Slaked lime and ehforine	. B)	Quick lime and chlorine		
C) Caleium and chlorine	D)	Bornt lime and chlorine		
EPAC name of K ₄ [Fe(CN) ₆] is:				
A) Potassium hexacyanoferrate (II)	B)	Potassium hexaeyanoferr	ate (III)
(1) Potassium hexacyanoiron (II)	D)	Patassium hexacyanoiron	(11)	
Which of the followings is not a conden-	sation pe	(ymer?		
A) Polystzrene B) Glyptal		C) Terelene	D)	Nylon-6,6

8. The proteins have their minimum solu	bility at :									
A) Acidic pH	B)	Basic pH								
C) Neutral pH	D)	Isoelectric point								
 On teating glucose with Fehling schu 	tion we get	a precipitate whose col	our is:							
A) Vellow B) Red		C) Black	D) White							
O. All muino acids in proteins are:										
A) Optically active except glycine	B)	Have L-configuration.								
(*) Roth A) and B)		Have D-configuration	The state of the s							
 Ethanol on reducing with NaBr give give rise to ally bromide. The function 	2 N. D. Supplied but the carry post-tants in presence of sulphunic acid									
A) Provide H*	B)	Convert OH of alcoho	of to a better leaving group water							
Cr Act as a dehydrating agent	(D)	All of these								
12. Edunot is made untit for drinking by	adding:									
A) Methanol B) Glycol		C) Glycerol	D) All of these							
3. A reimary amine can be distinguished	A grimary amine can be distinguished from a 2° or 3° amine by:									
A) Carbylamine reaction	B)	Reaction with CH ₃ 1								
(*) Reaction with acetyl obloride		None of these								
14 In Hinsberg test to distinguish between	In Hinsberg lest to distinguish between 1°, 2° and 3° arrine, the reagent used is :									
At SoClyHCl	B)	p-toluenesulphonyl ch	tortec							
Cit Palakania mid	D)	Benzenesulphoryl chi	oride							
 Chlorohenzene on reaction with a topyation of: 	nethyl chłor	ride in the presence of	anhydraus AICI ₃ results in the							
Ar Toluene		m-chiorestelatene								
C) m and p-chlorototuenes		Henzyl chloride								
46. Which of the following does not giv	e Cannizaer	e s reaction?								
A) HCHO	B)	2.2-dimethylprepaix								
C) Benzeldeliyde		2-methyl-2-phonyletl								
47. In the presence of iodine catalyst, c	blorine reac	ts with acetic acid to to	MITTE :							
AI CH3COCI	B)	CKIPGOGI								
C3 C3CH-COOH	D)	The state of the s								
48. Which of the following compounds	will be for	med by the reaction of l	IBr with acetylene?							
A' Filsylidene bromide	B	Ethy ene bromide								
gra Kida Phennich	D	Vinyl bromide	SHOW AND THE PARTY OF THE PARTY							
49. The organic reaction product from t	the reaction	of methyl magnesium l	bromide and ethyl alcohol is:							
At Methone B) Ethane		(') Propane	(D) Buranc							
50. Which of the following reagents ca	n distinguis	h C2H3OH from CH3O	H;							
A) H ₂ O B) NH ₃		C1 12+ KOH	D) IICI							

BIOLOGY

380	MOST theory reason for development or t	Committee	in moccio agai in penter	upa is .					
	A) Genetic recombination	B)	Aequired heritable change	je.					
	C) Random mutations	D)	Directed mutations						
52.	Test cross is crossing between								
	A Genotype with dominant trait	B)	Genotype with recessive	trait					
	Ct. F ₁ hybrid with double recessive	D)	Two F ₁ hybrids						
53.	Georetic maps of chromosomes are bas	sed on :							
	A) Non disjunction	100	Translocation						
	C) Dominarce	-	Genetic recombination						
54.	Grain colour of wheat is determined by resembling either parent in F2 generation		iirs of polygenes. In cross	SAABB	CC x asiblice, proget				
	Ag Half	1.018	One third						
	C) Less than 5%	11 95	75%	11. 1	did a secondor				
55.	Which mendelian idea is depicted by a			nbles bo	oth the parents :				
	A): Codominance		heemplete dominance						
	C) Law of dominance		Inheritance of one gene						
oft,	Which one of the following statements is not true for cancer cells in relation to mutations?								
	Mutations destroy telemerase inhibitor								
	Bit Mutations inactive the cell control								
	C) Mutations inhibit production of tele								
	By Mutations in proto-oneogenes acce				AND THE PARTY OF				
\$7.	The mechanism that causes a gene to m	iove from							
	A) Duplication B) Translocation		C) Crossing-over		Irversion				
58.	Number of triplet codous having all the	three ba	ses same in 64 triplet coo	lons is:					
	A) 12 B) 8		C) 6	D)					
59.	Which one of the following does not for	llow the	central dogma of molecul	ar biolog	N:				
	A) Mucor		Chlamydomonas						
	C) HIV		Pea						
60.	The unequivocal proof of DNA as the g								
-	At Virold B) Bacterial vir	FLIS	C) Bacterium	D)	Vangus				
61	Tess's of DNA fingerprenting is:								
	A) Relative proportion of purines and p								
	B) Relative difference in DNA occurre								
	(*) Relative amounts of DNA is ridges								
	D) Satellite DNA occurring as highly t	repeated	short DNA segments						

62.	Which one is not a	pplicable of RNA?							
	A) Complementary base pairing		B)	5' phosphoryl and 3' l	ydroxylo	ends			
	C) Beterocyclic n	itrogenous bases	D)	Chargaff's rule	Chargaff's rule				
				escape the heat for the d nely cole northern region					
	A): Western Grat			Meghalaya					
- 1	C) Corbett Nation	ral Park	D)	Keoladeo Ghana Natis	onal Park				
(14.	Tubeetomy:								
	A) Prevents implantation			Prevents foetal develo	ртен				
3	C) Prevents fertiliz	zation	D)	None of these					
65. I	Capacitaion of spe	rms occur in:							
- 1	 A) Female gerital 	tract	B)	Vas efferens					
3	Cr Vas Deferens		D)	Vagira					
	The deme is a group of:								
	Genes in different environment				Chromosomes in same organism				
. 1	C) Individuals in same environment		D)	Populations with same	Populations with same gene poot				
57.	Which one of the fo	allowing is the second	lary egg	membrane in mammals	:	DESIGNATION OF THE PERSON OF T			
7	A): Corona radiata		B)	Chorien					
_ 1	C) Vitelling memb	rane	D)	Zona pellucida					
8.	Anacmia in alcohol	lism may be due to the	e deficie	ency of :					
- 9	A) Yitamin H		B)	Vitamin B ₂					
3	() Falic acid & Vi	tamin B ₁₂	D)	Vitamin C					
19.	Mendel's experime	ntal organism was:							
1.3	A) Homo sapiens		B).	Antierbinum majus					
(C) Fisamsativum		D)	Drosophila melanogast	er				
70.	Arbelete foot disea	se is caused by:							
	A) Tinca pedis	B) Tinea capitis		C) Ricketssia	Acres of the	Candida afbicans			
1. 1	hi Frgeneration of	quantitative inheritan	ce, a rat	io of 1:4:6:4:1 is obtain	ed instead	lof:			
2	Ax 7.4:1:4	B) 3:1		C) 993:1	D)	8:6:4:1			
12.	The infective stage	of malaria is:							
. 7	A) Sporozoite	B) Merozoite		€) Schizont	D)	Gameles.			
13, 1	Who wrote the fam	ous book "Origin of	Species	The Real Bin					
- 2	A) Lamarek	B) Darwin		C) de Vries	D)	Mendel			
74. I	ha and food chain th	ne largest population i	is that o	f)					
1	A) Frimary consur	ners	B)	Tertiary consumers					
-	C) Froducers		D)	Decomposers					
75. 9	Smallest part of DN	A that undergoes rec	ombina	tion is:					
	A) Muton	B) Cistron		C) Replicon	D):	Recon			

OR MATHEMATICS

1	$Wa * h = a^{h-1} * bc a$	binary operation the	en 4 "	3 is equal to:		
	A) 16	B) 12		C) 64	D)	81
0.	Let A be the non-voic		n a fan	aity. The relation x is a broth	herse	fyon Ais:
	A) Reflexive	B) Symmetric		C) Transitive		None of these
3.	Range of the function	$f(x) = \frac{ x-1 }{ x-1 } \text{ is } t$				
	A) [-1.1]		13)	[-1,2]		
	C) [-22]		D)	None of these.		
				$ \tan^{-1} x - \inf + x \le 1$		
5 4 .	The domain of the de	rivative of the function	m fc	$y = \begin{cases} \frac{1}{2} (x - 1) & \text{if } x > 1 \end{cases}$	is:	
	A) IR-{G!			C) IR-(-1)	(D)	IR-1-1.11
95.	$\mathbf{H} A = \begin{bmatrix} \alpha & \beta \\ y & -\alpha \end{bmatrix} \text{ is s}$	with that $A^2 = 1$, then				
	A) $1 + \alpha^2 + \beta y = 0$	O .	13)	$1-\alpha^2+\beta\gamma=0$		
	(*) $1 - \alpha^2 - \beta y = 0$)	Di	$1+\alpha^2-B\gamma=0$		
Sti.	$\mathbb{R}^* A = \begin{bmatrix} 1 & 2 \\ 3 & 4 \end{bmatrix}. \text{ The}$					
	Cr. e7	Fc 1.3		$\begin{bmatrix} \frac{1}{2} & \frac{1}{3} \end{bmatrix}$		$\begin{bmatrix} 1 & 1 \\ -1 & -1 \end{bmatrix}$
	A) [0 1]	R) [-3 -4]		$C) \begin{bmatrix} \frac{1}{2} & -\frac{3}{3} \\ -\frac{1}{2} & 0 \end{bmatrix}$	D)	$\begin{bmatrix} \frac{1}{4} & \frac{1}{2} \\ \frac{1}{2} & 0 \end{bmatrix}$
	2'-1 23'	-1 4.5' -1				
57.	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	γ γ then t	he vuli	$\operatorname{neuf}\sum_{i,j}D_{i}$ is:		
	A) 0	 Β) αβγ 		C) 41+3+7	D	1 12 + 1/3"
58	The matrix $A = \begin{bmatrix} 0 \\ 1 \end{bmatrix}$	0 isa				
	A) Unknustrix		Bi	Diagonal matrix		
	C) Symmetric matri	X	13)	Skew-symmetrie manx		

59. If the function f is defined by f(x) =

B) 2

A) 1

 $\begin{cases} 3 & x \neq 0 \\ a+1 & x = 0 \end{cases}$ and f is continuous at x = 0, then value of [a, b]:

C) 3

=				
	If $f(x) = x-2 $ and $g(x) = f(f(x))$ then	for $x > 2$, g'(x) equals :	
	A) -1 B) 1		C) 0	D) None of these
	The normal to a given curve is parallel	to x-axis	if:	
	A) $\frac{dy}{dx} = 0$ B) $\frac{dy}{dx} = 1$		C) $\frac{dx}{dy} = 0$	D) $\frac{dx}{dy} = 1$
	The function $f(x) = \max\{(1-x), (1+x)\}$ A) Continuous at all points B) Differentiable at all points C) Differentiable at all points except a		(-∞,∞) is:	
	D) Continuous at all points except at x	≥ ± 1, w	here it is discontinuous	
	$\int \frac{dx}{e^x + e^{-x}}$ is equal to:			
-	A) $\tan^{-1}(e^{x}) + e^{-x}$	B)	$\tan^{-1}(e^{-x}) + c$	
	C) log (e ^x -e ^{-x}) + c		$\log (e^{v} + e^{-v}) + c$	
,	$\int_{(x-1)}^{x} \frac{dx}{(x-2)} $ equals:			
	A) $\log \left \frac{(x-1)^2}{x-2} \right + c$	B)	$\log \left \frac{(x-2)^2}{x-1} \right + c$	
	C) $\log \left \frac{(x-1)^2}{x-2} \right + c$	D)	$\log (x-1) (x-2) + c$	
	The area of the figure bounded by the	curves y		
	A) 2 B) 3		C) 4	D) 1
	Value of $\int \frac{e^{\epsilon}}{\sqrt{4-e^{2\epsilon}}} dx$ is:			
	A) $\sin^{-1}\left(\frac{e^{c}}{4}\right) + c$	B)	$\sin^{-1}(e^r)+e$	
	C) sin ⁻¹ (2e ⁺)+c	D)	$\sin^{-1}\left(\frac{e^{r}}{2}\right)+c$	
	Which of the following differential equa	tions has	$y = c_1 e^x + c_2 e^{-x}$ as the ger	neral solution:
	A) $\frac{d^2y}{dx^2} + y = 0$	B)	$\frac{d^2y}{dx^2} - y = 0$	
	C) $\frac{d^2y}{dx^2} + 1 = 0$	D)	$\frac{d^2y}{dx^2} - 1 = 0$	
	The general solution of the differential of	quation 2	$\frac{ydx - xdy}{y} = 0 \text{ is :}$	
	A) $xy = c$ B) $x = cy^2$		C) $y = cx$	D) $y = cx^2$

	2 to 1	diamber to the suscions Z - i	Find h = 1+ k is:
		dicentar to the vectors $\vec{a} = \vec{i}$ (C) 4	D) infinite
A) I	3) 2		
0. If a is a non-zero	vector of magnitude 'a an	d À is non-zero scaler, then	
	Dv · · · · ·	C) α= λ 	$D) \theta = \frac{1}{ X } *$
A) 1=1	B) λ = −[
It a. B. w are the	angles which a directed li	ne makes with the +ve din	ections of the co-ordinate axis
	$\beta + \sin^2 \gamma$ is equal to:		
A) I	B) 2	C) 3	D) None of these
77 The planes 2x –	y + 4z = 5 and $5x - 2.5y +$	10z = 6 are:	
A) Perpendicula		B) Parallel	
		(0.05	
C) Intersect y -	uxis	D) Passes through $\left(0,0,\frac{5}{4}\right)$	A CHARLES
73 The equation	$\int_{0}^{2} -2(\vec{r}' \cdot \vec{a}') + \lambda = 0 \text{ representation}$	sents a	
			D) None of these
A) Plane	B) Straight Line	C) Sphere	(a) thought or many
TE TE BY	By - O then Pt (/B) is:		
14. It P(A) = 2.1 (1	B = 0, then $P(A/B)$ is:		
	1	C: Not defined	D) 1
A) 0	B) 2	E: Mot actived	
75. HA and B are to	vo events, such that:		
$A'A \subseteq B$	b) B∈A	C) B = \$\phi\$	d) A = ø
27 1			
	GENARAL AWAR	RENESS OF THE SUBJE	CT
The State State (consider	of erop		
A STREET, SQUARE OF STREET, SQUARE, SQ	B) Wheat	C) Maize	D) Sarson
A) Grain			
	e pest of Ch	C) Maize	D) Grain
A) Wheat	B) Rice	THE RESERVE TO SERVE THE PARTY OF THE PARTY	
	ical substance which provid	C) Protection	D) Potassiure
A) Nitrogen	B) Energy	C) Tiblecani	
79. Flowering time		B) February- Mary	oh .
A) October- N	fovember_		
() June-July	or the contract of	D) August- Septen	iioci :
80is r	ative fruit plant of Amritser		D) Palm
A) Peach	B) Peas	C) Plum	D) rain
81. Best area for a	ople production is:	AND DESCRIPTION OF THE PERSON	TW Daniel
A) Maharasht		C) Kashmir	D) Punjab

82.	Seed rate of wheat po	eraere in kg is					
	A) 10 kg	B) 20 kg		C) .	30 kg	D)	40 kg
83.	Reason of low marke	a price of potato i	84				
	At More production		B)	Exper	t Problem		
	C; Bad quality produ		D)	All of	these		
	Sewing/planing time						Water State of the
	A) June-July	B) February-1	March	(C)	April-May	D)	August-September
85.	Low temperature dur	ing early style of	wheat is:				
. 100,000	A+ Beneficial		B)	l karant	iel)		
	C) Neither harmful r	e ther beneficial	D)	Nonz	of these		
\$6.	Water required to pro	oduce one kg of p	naddy is:				
	At 30 litres	B) 300 lines		C)	3000 litres	D)	30.000 fines
87	East- West length of	one acre is	Karm				
	A) 36	B) 40		C)	44	Dj	50
88.	Cotton sowing time i	in Punjab is					
	A) January-February	y			h-April		
	C) June-July		D)	Nove	mber-December		
89.	from the following v	which crops takes	more time	e to ma	iture (
	A) Toria	B) Season			Pea	D)	Barseem
90.	Fanting of potato is	done on					AH 255
	Ar Dry soi	B) Wet soil		(C)	Puddled soil	13)	All of these
91,	Which crop require !	ess no. of imigation	on :			-	2
	A) Wheat	B) Rice		()	Sugarcane	D).	Potato
02	Which one is Rabi C	Prop!?					
	A) Soybean	B) Lentil		C)	Arhar	D)	Cowpea
93.	From the following	which erop is used	d as fodde	7:	No tar Pin	CV	100
	A) Barley	B) Oat		(C)	Wheat	13	Rice
94.	Underground water	table on an avera	ge is		MADNIN	150	200 5 4
77.1	At 80 feet	B) 180 feet		0	280 feet	(3)	380 feet
93.	Whose milk contain	more fat :				70	West of the
	A) Cow	B) Buffide		C;	Goat	D)	Sheep
e)(x	Maida is the produc	d of				-	
	A) Rice	H) Maize		11313	Wheat	EF)	Grain
97	In the present period	d, one bag of whe	at in mkt	contr	ir Grams.	- 10	35
	A) 100 kg	B) 65 kg		C	50	4)	35
98.	Machine used to ha	rvest and thrash w	sheat is cr	fled:	Service management	Dec.	K
	A) Haramba	B) Thresher		CF	Combine harvester	(0)	Reaper
99	Loose smut (Kangia	an) is a disease of	*****			170	0.000
-	A) Rice	B) Wheat		(C)	Sunflower	(3)	Sugarcane
100	Geographical area of	of Punjab is:	. 1			1170	em (_ %
	A) 50 lac nector	B) 50 lac km	2	0	50 lae acres	(D)	50 kgc m ²