FACULTY OF SCIENCES

SYLLABUS FOR THE BATCH FROM THE YEAR 2024 TO YEAR 2028

Programme Code: BSHP

Programme Name

B.Sc. Physics (3 Years) / B.Sc. Physics (Honours) (4 Years)

(Semester I-II)

Examinations: 2024-25



Department of Physics

Khalsa College, Amritsar

(An Autonomous College)

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(b) Subject to change in the syllabi at any time.

(c) Please visit the college website time to time.

S.No.	PROGRAMME OBJECTIVES
1.	To develop human resource with knowledge, abilities and insight in Physics and
	Telated fields required for career in academia and industry.
2.	To apply principles of basic science concepts in understanding, analysis and prediction of physical systems.
3.	To understand the concepts and significance of the various physical phenomena.
4.	To carry out experiments to understand the laws and concepts of Physics.
5.	To acquire a wide range of problem solving skills and to apply them.

S.No.	PROGRAMME SPECIFIC OUTCOMES (PSOs)
PSO-1	To Demonstrate knowledge of classical mechanics, electromagnetism and modern physics and be able to apply this knowledge to analyse a variety of physical phenomena.
PSO-2	To demonstrate proficiency in mathematics and the mathematical concepts needed for a proper understanding of physics.
PSO-3	To demonstrate the ability to use skills in Physics and its related areas of technology for formulating and tackling Physics-related problems and identifying and applying appropriate physical principles and methodologies to solve a wide range of problems associated with Physics.
PSO-4	To analyse scientific reasoning for various things.
PSO-5	To enable students to be capable of analyzing the problems through different prospective and hence instill higher order thinking and learning.
PSO-6	To find ample career openings both in public as well as private sector enterprises. They can apply for all government jobs and find various opportunities in government organizations like DRDO, ISRO, BARC, IIT etc.

COURSE SCHEME											
SEMESTER-I											
Course	Course Title	Credits		Total Credits	Max. Ma		arks	Total Marks	Page No.	Syllabus Changed/	
Code		L	Т	Р		Th	Р	IA			Same as 2023- 24
]	Disc	iplin	e Specific	Cours	e (D	SC)			
BHP-111	MECHANICS-I	4	0	0	4	75		25	100	4-5	Same as 2023- 24
BHP -112	ELECTRICITY AND MAGNETISM-I	4	0	0	4	75		25	100	6-7	Same as 2023- 24
BHP -113	PHYSICS LAB-I	0	0	2	2		37	13	50	8-9	Same as 2023- 24
BHP -114	PHYSICS LAB-II	0	0	2	2		37	13	50	10	Same as 2023- 24
	Generic Elective (GE)										
MAP -111	MATHEMATICS –I	3	1	0	4	75		25	100	11-12	Same as 2023- 24
CHX -111	ORGANIC CHEMISTRY-I	3	0	0	3	56		19	75	13-14	Changed
CHP-112	ORGANIC CHEMISTRY LAB- I	0	0	1	1		19	06	25	15	Changed
		Α	bilit	y En	hancemen	t Cou	rse (A	AEC)			
BCEN-1123	COMMUNICATIVE ENGLISH-I	3	1	0	4	75		25	100	16-17	Changed
BHPB-1101	ਪੰਜਾਬੀ ਲਾਜ਼ਮੀ –।	4	0	0	4	75		25	100	18	Same as 2023- 24
BPBI-1102 BPHC- 1104	ਮੁੱਢਲੀ ਪੰਜਾਬੀ–I (In Lieu of Compulsory Punjabi) or PUNJAB HISTORY & CULTURE(For those students who are not domicile of Punjab)	4	0	0	4	75		25	100	19 20-21	Same as 2023- 24
			V	alue	Added Co	ourse (VAD)			
ZDA-111	DRUG ABUSE: PROBLEM, MANAGEMENT AND PREVENTION-I	1	0	0	1	25		-	25	22-23	Same as 2023- 24
			1	1	29		1		700		

SEMESTER-II												
Course	Course Title		Cred	its	Total Credits	Ma	ax. M	arks	Total Marks	Page No.	Syllabus Changed/ Same	
Code		L	Т	Р		Th	Р	IA			as 2023-24	
				I	Discipline	Specif	ic C	ourse (DSC)			
BHP-121	ELECTRICITY & MAGNETISM-II	4	0	0	4	75		25	100	24-25	Same as 2023- 24	
BHP-122	WAVES & OSCILLATIONS	4	0	0	4	75		25	100	26-27	Same as 2023- 24	
BHP -123	PHYSICS LAB-III	0	0	2	2		37	13	50	28-29	Same as 2023- 24	
BHP -124	PHYSICS LAB-IV	0	0	2	2		37	13	50	30-31	Same as 2023- 24	
	Generic Elective (GE)											
MAP -121	MATHEMATICS – II	3	1	0	4	75		25	100	32-33	Same as 2023- 24	
CHX -121	INORGANIC CHEMISTRY-II	3	0	0	3	56		19	75	34-35	Changed	
CHP-122	INORGANIC CHEMISTRY LAB- II	0	0	1	1		19	06	25	36-37	Changed	
		Α	bilit	y Enl	hancemen	t Cour	se (A	AEC)				
BCEN-1223	COMMUNICATIVE ENGLISH-II	3	1	0	4	75		25	100	38-39	Changed	
BHPB-1201	ਪੰਜਾਬੀ ਲਾਜ਼ਮੀ –II	4	0	0	4	75		25	100	40	Same as 2023- 24	
BPBI-1202	ਮੁੱਢਲੀ ਪੰਜਾਬੀ–II (In Lieu of Compulsory Punjabi)or PUNJAB		0		4	75		25	100	41	Same as 2023- 24	
BPHC-1204	HISTORY & CULTURE. (For those students who are not domicile of Punjab)	4	0	0	4	15				42-43		
	. u /		Va	alue	Added Co	urse ('	VAD)				
ZDA-121	DRUG ABUSE: PROBLEM, MANAGEMENT	1	0	0	1	25		-	25	44-45	Same as 2023- 24	
					29		<u> </u>		700			
									700			

SEMESTER I BHP-111 MECHANICS-I

Time: 3 Hours

400 Maximum Marks: 100 (Theory Marks: 75+Internal Assessment: 25) Pass Marks: 35%

Credits: L T P

Note for paper setter and students:

- 1. There will be five sections.
- 2. Section A is compulsory and will be of 15 marks consisting of 8 short answer type questions carrying 2.5 mark each covering the whole syllabus. The answer should not exceed 50 words. The candidate will have to attempt any 6 questions in this section.
- **3.** Sections B, C, D and E will be set from units I, II, III & IV respectively and will consist of two questions of 15 marks each from the respective unit. The candidates are required to attempt one question from each of these sections. Each question in these sections should not have more than two subparts.
- 4. Non-Programmable Scientific calculator is allowed.

Course Objectives: The purpose of the course is to provide the basic information about coordinate system and motion of particles in it, to understand the conservation laws and also to determine the difference between elastic and inelastic collisions. It includes applications of central force to the stability of circular orbits, Kepler's laws of planetary motion, orbital precession and Rutherford scattering, dynamics of rotating objects i.e. rigid bodies, angular velocity, the moment of inertia, the motion of rigid bodies and Euler equations. It also helps to understand the differences between types of forces and the inverse square force field.

Course Contents:

UNIT-I

Co-ordinate system and Motion of a Particle: Cartesian, **Plane polar** and spherical polar coordinate systems, area, volume, velocity and Acceleration in these systems. Solid angle, **Homogeneity of space and time, isotropy of space**, Symmetry principles and Laws of Conservation.

UNIT-II

Conservation of Momentum and Collisions: Internal forces and momentum conservation. Centre of mass, Elastic collisions in laboratory and center of mass systems; velocities, angles, energies in these systems and their relationships. Conservation of angular momentum and examples-shape of the galaxy, angular momentum of solar system, Torques due to internal forces, angular momentum about center of mass, Rutherford scattering, **Cross-section elastic scattering and impact parameter**.

UNIT-III

Inverse-Square-Law Force: Forces in nature (qualitative), Central forces, Potential energy and force between a point mass and spherical shell, a point mass and solid sphere, Two body problem and concept of reduced mass, Motion of a body under central force, equation of orbit in inverse-square force field, **Physical insight into the nature of motion**, Kepler's laws and their derivation, **Newton's law of gravitation from Kepler's law of planetary motion**.

UNIT-IV

Dynamics of Rigid Bodies: Rigid body, Moment of inertia of a rigid body, Equation of motion of a rigid body, rotational motion of a rigid body in general and in a plane lamina, Rotation of angular momentum vector about a fixed axis. Angular momentum and kinetic energy of a rigid body about principal axis, Euler's equations, Precession and elementary gyroscope, Spinning top.

Tutorial: Relevant problem on the topics covered in the course.

Books Prescribed:

- 1. Mechanics-Berkeley Physics Course, Vol-I (second edition):C. Kittel, W. D. Knight, M. A. Ruderman, C. A. Helmholtz and R. J. Moyer-Tata Mc Graw Hill Publishing Company Ltd., New Delhi.
- 2. Fundamentals of Physics: D. Halliday, R. Resnick and J. Walker (sixth edition)-Wiley India Pvt. Ltd.

Sr. No.	On completing the course, the students will be able to:
CO1	Understand the basic information about co-ordinate system and motion of particles in
	it.
CO2	Understand the conservation laws and also to determine the difference between elastic
	and inelastic collisions.
CO3	Explain the applications of central force to the stability of circular orbits, Kepler's laws
	of planetary motion, orbital precession and Rutherford scattering.
CO4	Understand the dynamics of rotating objects i.e. rigid bodies, angular velocity, the
	moment of inertia, the motion of rigid bodies and Euler equations.
CO5	Understand the differences between types of forces and also able to explain the inverse
	square force field.

Course Outcomes:

Time: 3 Hours

Credits: L T P 400 Maximum Marks: 100 (Theory Marks: 75+Internal Assessment: 25) Pass Marks: 35%

Note for paper setter and students:

- 1. There will be five sections.
- 2. Section A is compulsory and will be of 15 marks consisting of 8 short answer type questions carrying 2.5 mark each covering the whole syllabus. The answer should not exceed 50 words. The candidate will have to attempt any 6 questions in this section.
- **3.** Sections B, C, D and E will be set from units I, II, III & IV respectively and will consist of two questions of 15 marks each from the respective unit. The candidates are required to attempt one question from each of these sections. Each question in these sections should not have more than two subparts.
- 4. Non-Programmable Scientific calculator is allowed.

Course Objectives: Objectives of this course are to understand vectors calculus. To understand the fundamental laws and concepts in electricity and magnetism and their applications. To understand electric current and related concepts. To understand dielectrics, resistors, capacitors, and inductors. To understand the relativistic approach in electricity and magnetism.

Course Contents:

UNIT-I

Basic Ideas of Vector Calculus; Introduction to gradient, divergence & curl **and** their physical significance; **Laplacian and its formulation in rectangular, cylindrical and spherical coordinates systems**; Rules for vector derivatives; useful relations involving gradient, divergence & curl; Solenoidal and irrotational fields; Fundamental theorem for gradients-Gauss's and Stoke's theorems, Helmholz and Greens theorem.

UNIT-II

Electric charge and its properties; Coulombs law; Principal of Superposition; The electric field due to a point charge and continuous charge distributions; Electric field due to finite and infinite lines of charges; Field due to **sheet of charge**, electric dipole; **Electric** field lines **and** flux; Gauss's law; **differential form of Gauss's law**; applications of Gauss's law; Curl of electric field; **work and potential difference; potential difference as line integral of field**; Relation between potential and electric field; Poisson's and Laplace's equations; Electric potential due to different charge distribution-Wire, Ring, Disc, Spherical Sheet, Sphere, dipole etc.; The energy

for a point and continuous charge distribution.

UNIT-III

Concept of electrical images; Calculation of electric potential and field due to a point charge placed near an infinitely conducting sheet; Conductors in the electrostatic field; Capacitors; Current and current density; drift velocity; expression for current density vector; Equation of continuity; microscopic form of Ohm's law and electrical conductivity; limitations of Ohm's law; Multipoles and multipole moments; Equipotential surface method of electrical images.

UNIT-IV

Dielectrics, Non Polar and Polar Molecules, Polarisation of Dielectric, Polarization Vector, Atomic polarizability, Dielectric Constant, Capacity of a Capacitor, Electric Susceptibility, Free and Bound Charges, Gauss Law in Dielectric, and Displacement Vector, Energy stored in Capacitor having Dielectric Medium, Energy Density of a Dielectric Medium.

Tutorial: Relevant problem on the topics covered in the course.

Books Prescribed:

1. Introduction to Electrodynamics -D.J. Griffiths, Pearson Prentice Hall, New Delhi.

2. Electricity & Magnetism-T.S. Bhatia and Gurpreet Singh, Vishal Publications.

3. Berkeley Physics Course Vol. II (Electricity & Magnetism)- E.M.Purcell, Mc Graw hill, New York.

4. Fundamental of Physics -D. Halliday, R. Resnick and J. Walker (6th edition)-John Wiley, India Pvt. Ltd.

5. Electricity and Magnetism - A. K. Sikri, Pradeep Publications

Course Outcomes:

Sr. No.	On completing the course, the students will be able to:
CO1	Understand vectors calculus.
CO2	Understand the fundamental laws and concepts in electricity and magnetism and their
	applications.
CO3	Understand electric current and related concepts.
CO4	Understand dielectrics, resistors, capacitors, and inductors.
CO5	Understand the relativistic approach in electricity and magnetism.

SEMESTER I BHP-113 PHYSICS LAB-I

Credits: L T P

Time: 3 Hours

0 0 2 Maximum Marks: 50 (Max. Marks: 37+Internal Assessment: 13) Pass Marks: 35%

General Guidelines for Practical Examination

I. The distribution of marks is as follows:i) One experiment 15 Marksii) Brief Theory 5 Marks

iii) Viva–Voce 10Marks

iv) Record (Practical file) 7 Marks

II. There will be one sessions of 3 hours duration. The paper will have one session and will consist of 8 experiments out of which an examinee will mark 6 experiments and one of these is to be allotted by the external examiner.

III. Number of candidates in a group for practical examination should not exceed 12.

IV. In a single group no experiment be allotted to more than three examinee in any group.

Course Objectives: The purpose of the course isto understand the concept of moment of inertia, torque and angular acceleration with the help of fly wheel, conservation laws in elastic collision using one dimensional hanging spheres, modulus of rigidity, Young's modulus and Poisson's ratio.

LIST OF EXPERIMENTS

- 1. To measure the time periods of oscillation for the objects of various geometrical shapes but of same mass.
- 2. To study rotational motion using a flywheel and hence show that toque is proportional to angular acceleration.
- 3. To find the moment of inertia of an irregular body about an axis through its centre of gravity with a torsion pendulum.
- 4. To determine the moment of inertia of a flywheel.
- 5. To determine the Young's modulus by bending.
- 6. Determination of Poisson's ratio for rubber.
- 7. To verify laws of conservation of (a) linear momentum, (b) kinetic energy in elastic collisions using one dimensional collisions of hanging spheres. (c) Also determine energy transfer and coefficient of restitution.
- 8. To determine modulus of rigidity of copper wire by Maxwell needle experiment.

Books Prescribed

1. Practical Physics Vol. I, T.S. Bhatia, Gursharan Kaur, Iqbal Singh, Vishal Publications.

2. Practical Physics, C.L. Arora, S. Chand & Co.

Course Outcomes:

Sr. No.	On completing the course, the students will be able to:
CO1	Understand the concept of moment of inertia, torque and angular acceleration with the help of fly wheel.
CO2	Understand the conservation laws in elastic collision using one dimensional hanging sphere.
CO3	Determine modulus of rigidity of a wire
CO4	Determine Young's modulus of a beam.
CO5	Determine Poisson's ratio of rubber.

SEMESTER I BHP-114 PHYSICS LAB-II

Credits: L T P

002

Total Hours 60 Maximum Marks: 50 (Max. Marks: 37+Internal Assessment: 13) Pass Marks: 35%

General Guidelines for Practical Examination

I. The distribution of marks is as follows:

v)One experiment 15 Marks

vi) Brief Theory 5 Marks

Time: 3 Hours

vii) Viva–Voce 10Marks

viii) Record (Practical file) **7 Marks**

II. There will be one sessions of 3 hours duration. The paper will have one session and will consist of 8 experiments out of which an examinee will mark 6 experiments and one of these is to be allotted by the external examiner.

III. Number of candidates in a group for practical examination should not exceed 12.

IV. In a single group no experiment be allotted to more than three examinee in any group. **Course Objectives:** The purpose of the course is to understand the concept of charging and discharging of capacitor. It includes experiments to calculate low resistance with Carey Foster's Bridge and also to find out the capacitance and permittivity of a medium by de-Sauty's bridge. **LIST OF EXPERIMENTS**

- 1. To determine low resistance with Carey-Foster's Bridge.
- 2. To determine the resistance and specific resistance of a copper wire with the help of Kelvins double bridge.
- 3. To find the unknown capacitance of a capacitor by flashing and quenching of a neon lamp.
- 4. Measurement of capacitance, determination of permittivity of a medium air and relative permittivity by de-Sauty's bridge.
- 5. To study the variation in resistance of filament of a bulb with temperature.
- 6. To study the flow of water through a capillary tube as a function of pressure head using two tubes of same length but different radii.
- 7. To study the flow of water through a capillary tube as a function of pressure head using two tubes of different length but of same radii.

Books Prescribed

- 1. Practical Physics Vol. I, T.S. Bhatia, Gursharan Kaur, Iqbal Singh, Vishal Publications.
- 2. Practical Physics, C.L. Arora, S. Chand & Co.

Course Outcomes:

Sr. No.	On completing the course, the students will be able to:
CO1	Find out unknown capacitance.
CO2	Find out specific resistance of wire
CO3	To study the filament resistance.
CO4	Find out low resistance with Carey Foster's Bridge.
CO5	Find out the capacitance and permittivity of a medium by de-Sauty's bridge.

SEMESTER I MAP-111 MATHEMATICS –I

Time: 3 Hours

310 Maximum Marks: 100 (Theory Marks: 75+Internal Assessment: 25) Pass Marks: 35%

Credits: L T P

INSTRUCTIONS FOR PAPER SETTERS:

1. The question paper will consist of five sections namely Section-A which will be from entire syllabus (equally distributed from each unit) Section–B, C, D and E from Unit-I, II, III and IV, respectively.

2. The Section-A will consist of eight short answer type questions, each of 2.5 marks. Students are to attempt any six.

3. The Sections–B, C, D & E will consist of two questions each (each question should be subdivided into at most two parts). Students are to attempt any four questions in total by selecting one question from each section. Each question carries 15 marks.

4. Teaching time for this paper would be six periods per week.

Course Objectives:

• To acquire the knowledge of Functions and Derivatives.

• To have an idea about the Taylor and Maclaurin formulas.

• To get familiar with the concept of indefinite integrals.

Course contents:

UNIT-I

Functions and Derivatives: Limit, continuity and derivative of a function, geometrical significance of derivative, successive differentiation, Leibnitz theorem, maxima and minima of a function of single variable, partial derivatives, total derivative, chain rule.

UNIT-II

Differential Calculus: Rolle's theorem, mean value theorem, Taylor and Maclaurin formulas, Taylor series, concavity, point of inflexion, cusp point, asymptotes, graphs of standard planar curves in Cartesian coordinates.

UNIT-III

Anti derivatives: Indefinite integral as an anti derivative, method of substitution, partial fractions, integration by parts, reduction formulae.

Definite integrals: Definite integral as a limit of a sum, geometrical interpretation, double and triple integrals, applications of multiple integrals to determine centre of gravity and moments of inertia.

UNIT-IV

Matrices: Orthogonal matrices, Hermitian matrices, Unitary matrices; Cayley Hamilton theorem and its applications; rank of a matrix, consistency of a system of linear equations, eigen values and eigenvectors, diagonalization of matrices.

Reference Books:

1. Differential Calculus: Shanti Narayan, New Delhi, Shyam Lal, 1983.

2. Integral Calculus: Shanti Narayan, Delhi, S. Chand, 1968.

3. Higher Engineering Mathematics: B.S. Grewal, Delhi, Khanna, 1995.

COURSE OUTCOMES:

Sr. No.	On completing the course, the students will be able to:
CO1	Understand the concept of functions and derivatives.
CO2	Knowledge of differential calculus and its applications.
CO3	Apply concept of indefinite integrals in real life applications.

SEMESTER-I CHX 111 Organic Chemistry-I

Credits: L T P

Time: 3 Hours

3 00 Maximum Marks: 75 (Theory Marks: 56+Internal Assessment: 19) Pass Marks: 35%

INSTRUCTIONS FOR PAPER SETTERS AND CANDIDATES

(Scientific calculator is allowed)

- I. Examiner will make five sections of paper namely Section-I, II, III, IV and V
- II. Examiner will set total of NINE questions comprising ONE compulsory question of short answer type covering whole syllabi and TWO questions from each unit.
- III. Section-I will consist of eight six short questions carrying 2 marks each and students are required to attempt any six question.
- IV. Section-II, III, IV and V of paper will consist of EIGHT questions in total having TWO questions from each unit of the syllabus and each question carry 11 Marks.
- V. The students are required to attempt FIVE questions in all, taking ONE Compulsory question of section-I and one question from each section i.e. II, III, IV and V.

COURSE OBJECTIVES:

The objective of Organic Chemistry-I course is to enhance the knowledge of students on the topics of Stereochemistry especially in reference to the OPTICAL ISOMERISM. The course is also targeted to increase the knowledge of students for the various methods of preparation and properties of Alkanes, Alkenes, Alkynes, Arenes, aromaticity and Nucleophilic addition and Substitution reactions

COURSE CONTENTS:

UNIT-I

Stereochemistry: Molecular chirality, enantiomers/symmetry in achiral structures, chiral centres in chiral molecules, properties of chiral molecules-optical activity, absolute and relative configuration, the Cahn-Ingold Prelog R-S notional system physical properties of enantiomers. Resolution of enantiomers.

UNIT-II

Chemistry alkanes and alkenes: Conformations of alkanes and cycloalkanes: conformational analysis of ethane and n-butane; conformational analysis of cyclohexane, axial and equatorial bonds. Difference between configuration and conformation. Stereochemistry of alkenes. Mechanism of hydrogenation of alkenes, stereochemistry of hydrogenation of alkenes, Dehydration of alcohols and regioselectivity of these reactions. Acid catalysed dehydration of alcohols with complete mechanistic discussion,

Alkynes: Acidity of acetylene and terminal alkenes, metal ammonia reduction of alkyne, addition of hydrogen halides and water to alkynes with mechanism.

UNIT-III.

15Hrs

Nucleophilic substitution and addition reaction:

(a) Mechanism of nucleophilic substitution (SN^1/SN^2) , stereochemistry of SN^1/SN^2 reactions, carbocation stability and the rate of substitution by the SN^1 mechanism, stereochemistry of SN^1 reactions, carbocation rearrangements in SN^1 reactions, solvent effects, substitution and elimination as competing reactions.

(b) Principles of nucleophillic addition to carbonyl groups : Hydration, acetal formation, cyanohydrin

15Hrs in chir

15Hrs

formation, reactions with primary and secondary amines, Wittig reaction, steroselective addition to carbonyl groups, mechanism of halogenation and aldol condensation

UNIT-IV

15Hrs

Arenes and Aromaticity

Structure of benzene and its derivatives, molecular formula and kekule structure. Stability and resonance structure of benzene with its MO picture. Aromaticity: the Huckel's rule, aromatic ions.

Aromatic electrophilic substitution-general pattern of the mechanism, role of σ and π complexes. Mechanism of nitration, halogenation, sulphonation, mercuration and Friedel Crafts reaction. Energy profile diagrams. Activating and deactivating substituents, orientation and ortho/pararatio.

BOOKS PRESCRIBED

- 1. R.T. Morison and R.N. Boyd, Organic Chemistry.
- 2. I.L. Finar, Organic Chemistry, Vol. I IV ed.
- 3. Advanced Organic Chemistry, Reactions Mechanisms and Structure by J. March.
- 4. Schaum's Outlines Series Theory and Problems of Organic Chemistry by Herbert Meislick and Jacob Sharefkin
- 5. Problems and their solution in Organic chemistry by I.L. Finar, Modern Organic Chemistryby J.D. Robbert and M.C. Caserio.
- 6. Organic Chemistry by D.J. Cram and G.S. Hammond.
- 7. J.E. Banks, Naming Organic Compounds Programmed Introduction to Organic Chemistry.
- 8. E.L. Eliel, Stereochemistry of carbon compounds.
- 9. W. Camp, Organic Spectroscopy.
- 10. F.A. Carey, Organic Chemistry.

COURSE OUTCOMES:

S. No.	On completing the course, the outcomes will be as under:
CO1	Learned about SN1, SN2 and SNi Mechanism and the related stereochemistry.
CO2	Understand the concept, principle and applications of UV, IR and NMR Spectroscopy and the problems pertaining to the structure elucidation of simple organic compounds.
CO3	Solved the elimination reaction problems
CO4	Distinguish between type of addition, elimination and substitution reaction.
CO5	Learn E and Z nomenclature ,Stereo chemical principal, enantiomeric relationship R and S

SEMESTER-I CHP-112 Organic Chemistry Lab-I

Time: 3 Hours

Credits: L T P 0 0 1 Maximum Marks: 25 (Max. Marks: 19+Internal Assessment: 06) Pass Marks: 35%

INSTRUCTIONS FOR PAPER SETTERS AND CANDIDATES:

I. Examiner will give one organic salt to the students.

II. Each student will get different salt and analyse it for elements, functional group and prepare its derivatives.

III. The question paper will be 19 marks with split as under:

(Write up = 6, Performance = 6, Viva-voce = 5, Practical note book = 2)

COURSE OBJECTIVES:

In organic chemistry practical students will learn about the Evaluation of organic compounds for the detection of element, functional group and preparation of their derivatives. It includes following functional groups: Acids, ketones, aldehyde, carbohydrates, aromatic hydrocarbons, aromatic amines and phenols.

COURSE CONTENTS:

The preliminary examination of physical and chemical characteristics (physical state, colour, odour and ignition tests), elemental analysis (nitrogen, sulphur, chlorine, bromine, and iodine), and solubility tests including acid-base reactions, classification tests involving functional reactivity other than acid-base test, preparation of derivatives for given pure organic compounds.

The following categories of compounds should be analyzed.

-phenols, carboxylic acids

-carbonyl compounds - ketones, aldehydes

-carbohydrates

-aromatic amines

-aromatic hydrocarbons

BOOKS PRESCRIBED:

Practical Organic Chemistry by F.G. Mann and B.C. Saunders

COURSE OUTCOMES:

S. No.	On completing the course, the student will be able to
CO1	Perform functional group analysis
CO2	Prepare derivatives of organic compounds
CO3	Determine of physical constant: Melting point, Boiling point.
CO4	Learn different separation techniques.
CO5	Learn how to perform TLC

SEMESTER – I BCEN-1123

COMMUNICATIVE ENGLISH -I

B.Sc. (Hons.) Physics, Chemistry, Botany, Zoology, Maths and Computational Statistics and Data

Analytics

Credits: L T P

310

Time: 3 Hours Maximum marks 100 (Theory 60 Marks + Practical 15 Marks+ Internal Assessment 25 Marks)

Pass Marks: 35%

Instructions for the Paper Setter and Distribution of Marks:

The question paper will consist of four sections and the distribution of marks will be as under: Section A: 12 Marks Section B: 12 Marks Section C: 18 Marks Section D: 18 Marks

Section A

1. Fifteen (15) Questions on the usage of Preposition, Articles, and Change of Voice will be set. The students will be required to attempt any Twelve (12).

(12X1=12 Marks)

Section B

2. ONE question (with sub parts) based on Skills and Strategies development exercises in Unit-1 and Unit-2 of the prescribed text book *Making Connections* will be set.

(1X12= 12marks)

Section C

3. Five short answer type questions from Unit 1 and 2 of *Making Connections : A Strategic Approach To Academic Reading* will be set. The students will be required to attempt any three.

(3X2=06 marks)

4. Four Essay type question (Two from each unit) from Unit 1 and 2 of *Making Connections: A strategic Approach to Academic Reading* will be set. The students will be required to answer any two, choosing at least one from each unit.

(2X6= 12 marks)

(1X6 = 6 marks)

Section D

5. A Comprehension questions of an unseen passage will be set. (1X6 = 6 marks)6. A question requiring the students to write a Paragraph on ONE of the TWO given topics.

(1X6 = 6 marks)

7. A question requiring the students to write an Official/Business Letter on ONE of the TWO given Topics.

Course Objectives:

I: To develop competence in written communication.

II: To inculcate innovative and critical thinking among the students.

III: To enable them to grasp the application of communication theories.

IV: To acquire the knowledge of latest technology related with communication skills.

V: To provide knowledge of multifarious opportunities in the field of this programme.

Course Contents:

1. Reading and Comprehension Skills:

(a) Reading tactics and strategies; Reading purposes-kinds of purposes and associated comprehension; Reading for direct meanings.

(b) The Students will be required to read and comprehend the essays in Unit 1 and 2 of the book *Making Connections: A Strategic Approach to Academic Reading* by Kenneth J. Pakenham, Third Edition.

2. Writing Skills: Guidelines for effective writing; writing styles for paragraph and official/ business letter.

3. Grammar: Preposition, Articles, and Change of Voice.

Prescribed Books:

Making Connections by Kenneth J. Pakenham 3rd Edn. CUP

Recommended Books:

- 1. Oxford Guide to Effective Writing and Speaking by John Seely.
- 2. The Written Word by Vandana R Singh, Oxford University Press

Course Outcomes:

The completion of this course enables students to:

- 1. Identify common errors in language and rectify them.
- 2. Develop and expand writing skills through controlled and guided activities.

3. Develop coherence, cohesion and competence in written discourse through intelligible pronunciation.

4. Develop the ability to handle the interview process confidently and learn the subtle nuances of an effective group discourse.

5. Communicate contextually in specific and professional situations with courtesy.

PRACTICAL (Marks: 15)

Course Contents:-

- 1. Reading dialogues (5 Marks)
- 2. Rapid reading (5 Marks)
- 3. Project File (5 Marks)

SEMESTER – I

BHPB-1101 Punjabi (Compulsory)-1 ਪੰਜਾਬੀ (ਲਾਜ਼ਮੀ)−1

Credit & Marks Distribution, Eligibility and Pre-Requisites of the Course

Course title & Code	Total Teaching Hours	Total Credits/ Hours	di	Cred stribu	it tion	Total M 100	larks)	Time Allowed in Exam	Eligibility criteria	Pre- requisite of the course (if any)
	nours	per week	L	Т	Р	Theory	IA			
ਪੰਜਾਬੀ ਲਾਜ਼ਮੀ)–1 BHPB-1101	60	4	4	0	0	75	25	3 Hours	Class 12th pass in any stream	Studied Punjabi up to 10th Standard

ਕੋਰਸ ਦਾ ਉਦੇਸ਼ Course Objective	ਪਾठ-ਕ੍ਰਮ ਨਤੀਜੇ Course Outcomes (COs)			
 ਵਿਦਿਆਰਥੀਆਂ ਵਿਚ ਸਾਹਿਤਕ ਰੁਚੀਆਂ ਪੈਦਾ 	 ਉਸ ਵਿਚ ਸਾਹਿਤ ਰੁਚੀਆਂ ਵਿਕਸਤ ਹੋਣਗੀਆਂ। 			
ਕਰਨਾ।	 ਉਸ ਵਿਚ ਸਾਹਿਤ ਸਿਰਜਣਾ ਦੀ ਸੰਭਾਵਨਾ ਵਧੇਗੀ। 			
 ਆਲੋਚਨਾਤਮਕ ਰਚੀਆਂ ਵਿਕਸਤ ਕਰਨਾ। 	 ਉਸ ਵਿਚ ਕਿਸੇ ਵੀ ਵਿਸ਼ੇ ਦਾ ਗਹਿਨ ਅਧਿਐਨ ਕਰਨ ਦਾ ਬੋਧ ਹੋਵੇਗਾ। 			
 ਮਾਤ ਭਾਸ਼ਾ ਦੀ ਸਮਝ ਨੂੰ ਵਿਕਸਤ ਕਰਨਾ। 	 ਉਹ ਪੰਜਾਬੀ ਭਾਸ਼ਾ ਦੇ ਨਿਕਾਸ ਤੇ ਵਿਕਾਸ ਬਾਰੇ ਗਿਆਨ ਹਾਸਲ ਕਰਨਗੇ 			

ਅੰਕ–ਵੰਡ ਅਤੇ ਪ੍ਰੀਖਿਅਕ ਲਈ ਹਦਾਇਤਾਂ

ਸਿਲੇਬਸ ਦੇ ਚਾਰ ਭਾਗ ਹਨ ਪਰ ਪ੍ਰਸ਼ਨ-ਪੱਤਰ ਦੇ ਪੰਜ ਭਾਗ ਹੋਣਗੇ। ਪਹਿਲੇ ਭਾਗ ਵਿਚ 1.5-1.5 (ਡੇਢ-ਡੇਢ) ਅੰਕ ਦੇ ਅਤਿ-ਸੰਖੇਪ (Objective Type) 10 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ ਜੋ ਕਿ ਸਾਰੇ ਸਿਲੇਬਸ ਵਿਚੋਂ ਹੋਣਗੇ ਅਤੇ ਸਾਰੇ ਪ੍ਰਸ਼ਨ ਹੱਲ ਕਰਨੇ ਲਾਜ਼ਮੀ ਹੋਣਗੇ। ਸਿਲੇਬਸ ਦੇ ਬਾਕੀ ਚਾਰ ਭਾਗਾਂ ਵਿਚ 02-02 ਲੇਖ ਨੁਮਾ ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ। ਹਰੇਕ ਭਾਗ ਵਿਚੋਂ 01-01 ਪ੍ਰਸ਼ਨ ਕਰਨਾ ਲਾਜ਼ਮੀ ਹੋਵੇਗਾ। ਹਰੇਕ ਪ੍ਰਸ਼ਨ ਦੇ ਬਰਾਬਰ 15 ਅੰਕ ਹੋਣਗੇ। ਪੇਪਰ ਸੈੱਟਰ ਜੇਕਰ ਚਾਹੇ ਤਾਂ ਪ੍ਰਸ਼ਨਾਂ ਦੀ ਵੰਡ ਅੱਗੋਂ ਵੱਧ ਤੋਂ ਵੱਧ ਚਾਰ ਉਪ-ਪ੍ਰਸ਼ਨਾਂ ਵਿਚ ਕਰ ਸਕਦਾ ਹੈ।

ਪਾਠ–ਕ੍ਰਮ

ਭਾਗ–ਪਹਿਲਾ

ਕਾਵਿ ਕਥਾ, (ਕਵਿਤਾ ਅਤੇ ਕਹਾਣੀ) ਡਾ. ਮਹਿਲ ਸਿੰਘ (ਮੁੱਖ ਸੰਪਾਦਕ) ਅਤੇ ਡਾ. ਆਤਮ ਸਿੰਘ ਰੰਧਾਵਾ (ਸੰਪਾਦਕ), ਕਸਤੂਰੀ ਲਾਲ ਐਂਡ ਸਨਜ਼, ਅੰਮ੍ਰਿਤਸਰ।

(ਕਵਿਤਾ ਭਾਗ ਵਿਚੋਂ ਪ੍ਰਸੰਗ ਸਹਿਤ ਵਿਆਖਿਆ/ਕਵਿਤਾ ਦਾ ਵਿਸ਼ਾ-ਵਸਤੁ। ਕਹਾਣੀ ਭਾਗ ਵਿਚੋਂ ਸਾਰ/ਵਿਸ਼ਾ-ਵਸਤੁ)

ਭਾਗ–ਦੁਜਾ

ਪੰਜਾਬ ਦੇ ਮਹਾਨ ਕਲਾਕਾਰ (ਬਲਵੰਤ ਗਾਰਗੀ) ਗੁਰੂ ਨਾਨਕ ਦੇਵ ਯੂਨੀਵਰਸਿਟੀ, ਅੰਮ੍ਰਿਤਸਰ। (ਅੰਮ੍ਰਿਤਾ ਸ਼ੇਰਗਿੱਲ ਤੋਂ ਭਾਈ ਸਮੁੰਦ ਸਿੰਘ ਤਕ) (ਵਿਸ਼ਾ-ਵਸਤ/ਸਾਰ/ਨਾਇਕ ਬਿੰਬ)

ਭਾਗ–ਤੀਜਾ

(ੳ) ਪੈਰ੍ਹਾ ਰਚਨਾ (ਤਿੰਨਾਂ ਵਿਚੋਂ ਇਕ)

(ਅ) ਪੈਰ੍ਹਾ ਪੜ੍ਹ ਕੇ ਪ੍ਰਸ਼ਨਾਂ ਦੇ ਉੱਤਰ

ਭਾਗ–ਚੌਥਾ

(ੳ) ਭਾਸ਼ਾ ਵੰਨਗੀਆਂ: ਭਾਸ਼ਾ ਦਾ ਟਕਸਾਲੀ ਰੂਪ, ਭਾਸ਼ਾ ਅਤੇ ਉਪ-ਭਾਸ਼ਾ ਵਿਚਲਾ ਅੰਤਰ, ਪੰਜਾਬੀ ਉਪ-ਭਾਸ਼ਾਵਾਂ ਦੇ ਪਛਾਣ-ਚਿੰਨ੍ਹ।
(ਅ) ਪੰਜਾਬੀ ਭਾਸ਼ਾ: ਨਿਕਾਸ ਤੇ ਵਿਕਾਸ।

SEMESTER – I BPBI-1102 Basic Punjabi -1 ਮੁਢਲੀ ਪੰਜਾਬੀ–1

(In Lieu of Compulsory Punjabi)

Credit & Marks Distribution, Eligibility and Pre-Requisites of the Course

Course title &Code	Total Teaching Hours	Total Credits/	Credit distribution		Total Marks 100		Time Allowed in Exam	Eligibility criteria	Pre- requisite of the course (if any)	
	nours	per week	L	Т	Р	Theory	IA			
ਮੁਢਲੀ ਪੰਜਾਬੀ−1 BPBI-1102	60	4	4	0	0	75	25	3 Hours	Class 12th pass in any stream	NOT Studied Punjabi up to 10th Standard

ਕੋਰਸ ਦਾ ਉਦੇਸ਼ Course Objective	ਪਾਠ-ਕ੍ਰਮ ਨਤੀਜੇ Course Outcomes (COs)
 ਵਿਦਿਆਰਥੀ ਨੂੰ ਗੁਰਮੁਖੀ ਲਿਪੀ ਤੋਂ ਜਾਣੂ ਕਰਾਉਣਾ। 	• ਵਿਦਿਆਰਥੀ ਪੰਜਾਬੀ ਭਾਸ਼ਾ ਅਤੇ ਗੁਰਮੁਖੀ ਲਿਪੀ ਦੀ ਸਿਖਲਾਈ ਵਿਚ ਮੁਹਾਰਤ
 ਵਿਦਿਆਰਥੀ ਨੂੰ ਸ਼ੁੱਧ ਪੰਜਾਬੀ ਪੜ੍ਹਨਾ-ਲਿੱਖਣਾ ਸਿਖਾਉਣਾ। 	ਹਾਸਲ ਕਰਨਗੇ।
 ਪੰਜਾਬੀ ਭਾਸ਼ਾ ਦੀਆਂ ਵਿਆਕਰਨਕ ਬਾਰੀਕੀਆਂ ਤੋਂ ਜਾਣੂ 	• ਪੰਜਾਬੀ ਭਾਸ਼ਾ ਵਿਚ ਮੁਹਾਰਨੀ, ਲਗਾਂ-ਮਾਤਰਾਂ, ਸਵਰ ਅਤੇ ਵਿਅੰਜਨ ਅੱਖਰਾਂ ਦੀ
ਕਰਾਉਣਾ।	ਪਛਾਣ ਅਤੇ ਵਰਤੋਂ ਸਬੰਧੀ ਸਮਝ ਵਿਕਸਿਤ ਹੋਵੇਗੀ।
 ਸ਼ੁੱਧ ਸੰਚਾਰ ਨੂੰ ਵਿਕਸਤ ਕਰਨਾ। 	 ਵਿਦਿਆਰਥੀ ਸ਼ੁੱਧ ਪੰਜਾਬੀ ਲਿਖਣ-ਪੜ੍ਹਨ ਦੇ ਸਮਰੱਥ ਹੋਣਗੇ।
	 ਉਹ ਪੰਜਾਬੀ ਭਾਸ਼ਾ ਦੇ ਸ਼ੁੱਧ ਰੂਪਾਂ ਦੀ ਜਾਣਕਾਰੀ ਹਾਸਲ ਕਰਨਗੇ।

ਅੰਕ-ਵੰਡ ਅਤੇ ਪ੍ਰੀਖਿਅਕ ਲਈ ਹਦਾਇਤਾਂ

ਸਿਲੇਬਸ ਦੇ ਚਾਰ ਭਾਗ ਹਨ ਪਰ ਪ੍ਰਸ਼ਨ-ਪੱਤਰ ਦੇ ਪੰਜ ਭਾਗ ਹੋਣਗੇ। ਪਹਿਲੇ ਭਾਗ ਵਿਚ 01-01 ਅੰਕ ਦੇ ਅਤਿ-ਸੰਖੇਪ ਉੱਤਰ ਵਾਲੇ (Objective Type) 11 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ ਜੋ ਕਿ ਸਾਰੇ ਸਿਲੇਬਸ ਵਿਚੋਂ ਹੋਣਗੇ ਅਤੇ ਸਾਰੇ ਪ੍ਰਸ਼ਨ ਹੱਲ ਕਰਨੇ ਲਾਜ਼ਮੀ ਹੋਣਗੇ।ਪ੍ਰਸ਼ਨ ਪੱਤਰ ਦੇ ਦੂਸਰੇ ਅਤੇ ਤੀਸਰੇ ਭਾਗ ਵਿਚ, ਸਿਲੇਬਸ ਦੇ ਪਹਿਲੇ ਅਤੇ ਦੂਸਰੇ ਭਾਗ ਵਿਚੋਂ 8-8 ਅੰਕਾਂ ਦੇ 3-3 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ। ਜਿੰਨ੍ਹਾਂ ਵਿਚੋਂ ਵਿਦਿਆਰਥੀ ਨੇ ਕੋਈ 2-2 ਪ੍ਰਸ਼ਨ ਹੱਲ ਕਰਨੇ ਹੋਣਗੇ। ਇਸੇ ਤਰ੍ਹਾਂ ਪ੍ਰਸ਼ਨ ਪੱਤਰ ਦੇ ਚੌਥੇ ਭਾਗ ਵਿਚ 4-4 ਅੰਕਾਂ ਦੇ 5 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ। ਜਿੰਨ੍ਹਾਂ ਵਿਚੋਂ ਵਿਦਿਆਰਥੀ ਨੇ ਕੋਈ 2-2 ਪ੍ਰਸ਼ਨ ਹੱਲ ਕਰਨੇ ਹੋਣਗੇ।ਭਾਗ ਪੰਜਵੇਂ ਵਿਚ 2-2 ਅੰਕਾਂ ਦੇ 10 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ। ਜਿੰਨ੍ਹਾਂ ਵਿਚੋਂ ਵਿਦਿਆਰਥੀ ਨੇ 8 ਪ੍ਰਸ਼ਨ ਕਰਨੇ ਲਾਜ਼ਮੀ ਹੋਣਗੇ।ਭਾਗ ਪੰਜਵੇਂ ਵਿਚ 2-2 ਅੰਕਾਂ ਦੇ 10 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ। ਜਿੰਨ੍ਹਾਂ ਵਿਚੋਂ ਵਿਦਿਆਰਥੀ ਨੇ 8 ਪ੍ਰਸ਼ਨ ਕਰਨੇ ਲਾਜ਼ਮੀ ਹੋਣਗੇ।

ਪਾਠ–ਕ੍ਰਮ ਭਾਗ–ਪਹਿਲਾ

(ੳ) ਪੰਜਾਬੀ ਭਾਸ਼ਾ ਤੇ ਗਰਮਖੀ ਲਿਪੀ:

ਨਾਮਕਰਣ ਤੇ ਸੰਖੇਪ ਜਾਣ-ਪਛਾਣ: ਗੁਰਮੁਖੀ ਵਰਣਮਾਲਾ, ਅੱਖਰ ਕ੍ਰਮ, ਸਵਰ ਵਾਹਕ (ੳ, ਅ, ੲ), ਲਗਾਂ-ਮਾਤਰਾਂ, ਪੈਰ ਵਿਚ ਬਿੰਦੀ ਵਾਲੇ ਵਰਨ, ਪੈਰ ਵਿਚ ਪੈਣ ਵਾਲੇ ਵਰਨ, ਬਿੰਦੀ, ਟਿੱਪੀ, ਅੱਧਕ

(ਅ) ਸਿਖਲਾਈ ਤੇ ਅਭਿਆਸ

ਭਾਗ–ਦੂਜਾ

ਗੁਰਮੁਖੀ ਆਰਥੋਗਰਾਫੀ ਅਤੇ ਉਚਾਰਨ:

ਸਵਰ, ਵਿਅੰਜਨ: ਮੁਢਲੀ ਜਾਣ-ਪਛਾਣ ਅਤੇ ਉਚਾਰਨ, ਮੁਹਾਰਨੀ, ਲਗਾਂ-ਮਾਤਰਾਂ ਦੀ ਪਛਾਣ

ਭਾਗ–ਤੀਜਾ

ਪੰਜਾਬੀ ਸ਼ਬਦ-ਜੋੜ: ਮੁਕਤਾ (ਦੋ ਅੱਖਰਾਂ ਵਾਲੇ ਸ਼ਬਦ, ਤਿੰਨ ਅੱਖਰਾਂ ਵਾਲੇ ਸ਼ਬਦ), ਸਿਹਾਰੀ ਵਾਲੇ ਸ਼ਬਦ, ਬਿਹਾਰੀ ਵਾਲੇ ਸ਼ਬਦ, ਔਂਕੜ ਵਾਲੇ ਸ਼ਬਦ, ਦੁਲੈਂਕੜ ਵਾਲੇ ਸ਼ਬਦ, ਲਾਂ ਵਾਲੇ ਸ਼ਬਦ, ਦੁਲਾਵਾਂ ਵਾਲੇ ਸ਼ਬਦ, ਹੋੜੇ ਵਾਲੇ ਸ਼ਬਦ, ਕਨੌੜੇ ਵਾਲੇ ਸ਼ਬਦ, ਲਗਾਖਰ (ਬਿੰਦੀ, ਟਿੱਪੀ, ਅੱਧਕ) ਵਾਲੇ ਸ਼ਬਦ

ਭਾਗ–ਚੌਥਾ

ਸ਼ੁੱਧ-ਅਸ਼ੁੱਧ ਸ਼ਬਦ

SEMESTER – I BPHC-1104 PUNJAB HISTORY & CULTURE (From Earliest Times to C 320 BC)

(Special Paper in lieu of Punjabi compulsory) (For those students who are not domicile of Punjab)

BA, B.A. SS/ B. A. (Hons. – English), B. Com. (Hons., Regular, Account. & Finance), B. Sc. Bio-Tech./Comp. Sc./Eco./Fashion Designing/Food Science/IT/Med./Non Med., B.Sc. (Hons. –Botany, Chemistry, Mathematics, Physics, Zoology), i B. of Mult.; B. in Int. & Mob. Tech.; BBA; BCA; BJMC; B. Voc. (Software Development, Theatre and Stage Craft, FoodProcessing, Textile Design & Apparel Technology) BTTM (Bachelor of Tour and Travel Management)

> Credits: L T P 400 Maximum Marks: 100 (Theory Marks: 75+Internal Assessment: 25) Pass Marks: 35%

Time: 3 Hours

Instructions for the Paper Setters:

Question paper should consist of two sections—Section A and Section B. The paper setter must ensure that questions in Section–A do not cover more than one point, and questions in Section–B should cover at least 50 per cent of the theme.

Section–A: The examiner will set 15 objective type questions out of which the candidate shall attempt any 10 questions, each carrying $1\frac{1}{2}$ marks. The total weightage of this section will be 15 marks. Answer to each question should be in approximately one to two sentences.

Section–B: The examiner will set 8 questions, two from each Unit. The candidate will attempt 4 questions selecting one from each Unit in about 1000 words. Each question will carry 15 marks. The total weightage of this section will be 60 marks.

Note: The examiner is to set the question paper in two languages: English & Hindi.

Course Objectives: The main objective of this course is to educate the history and culture of the Ancient Punjab to the students who are not domicile of the Punjab. It aims to familiarize these students with the physical features of ancient Punjab and its impact on its history and culture. It also provides them information about the different sources to construct the history and culture of the ancient Punjab. The course intends to provide knowledge of social, economic, religious life of the Harappan civilization, Indo-Aryans, teachings and impact of Jainism and Buddhism in the Punjab.

Unit-I

- 1. Physical features of the Punjab and impact on history.
- 2. Sources of the ancient history of Punjab.

Unit-II

- 3. Harappan Civilization: Town planning; social, economic and religious life of the Indus Valley People.
- 4. The Indo-Aryans: Original home and settlement in Punjab.

Unit-III

- 5. Social, Religious and Economic life during Rig Vedic Age.
- 6. Social, Religious and Economic life during later Vedic Age.

Unit-IV

- 7. Jainism in the Punjab.
- 8. Teachings and impact of Buddhism.

Suggested Readings:-

L. Joshi (ed), History and Culture of the Punjab, Art-I, Patiala, 1989 (3rd edition)

L.M. Joshi and Fauja Singh (ed), History of Punjab, Vol. I, Patiala 1977.

Budha Parkash, Glimpses of Ancient Punjab, Patiala, 1983.

B.N. Sharma, Life in Northern India, Delhi. 1966.

Course Outcomes:

On Completing the Course, the Students will be able to :

- **CO-1** Learn the history and culture of the Ancient Punjab.
- CO-2 Study the physical features of ancient Punjab.
- **CO-3** Understand about the sources of the history of the Punjab.
- **CO-4** Analyse the social, economic, religious life of the Harappan civilization and Vedic-Aryans.
- CO-5 Learn the teachings and impact of Jainism and Buddhism in the Punjab.

SEMESTER – I Course Code: ZDA111 Course Title- Drug Abuse: Problem, Management and Prevention-I (Compulsory for all Under Graduate Classes)

Time: 3 Hours

Credits: L T P 100 Maximum Marks: 25 Pass Marks: 35%

Instructions for the Paper Setters:

- 1) There will be two sections A and B.
- 2) Section A is compulsory and will be of 5 marks consisting of 8 short answer type questions carrying 1 marks each covering the whole syllabus. The candidates are required to attempt 5 questions out of 8 short answer type questions. The answer should not exceed 50 words.
- 3) Candidates shall be required to attempt 4 questions from Section B, selecting one question from each unit and each question carries 5 marks. Preferably, the question should not be split into more than two sub-parts.

Course Objectives- The course aims to:

CO-1.	Generate the awareness against drug abuse.
CO-2.	Describe a variety of models and theories of addiction and other problems
	related to substance abuse.
CO-3.	Describe the behavioral, psychological, physical health and social impact of
	psychoactive substances.
CO-4.	Provide culturally relevant formal and informal education programs that
	raise.awareness and support for substance abuse prevention and the recovery process.
CO-5.	Describe factors that increase likelihood for an individual, community or group.to
	be at risk of substance use disorders.

Unit-I

• Meaning of Drug Abuse

Meaning of drug abuse Nature and Extent of Drug Abuse: State and National Scenario

UNIT-II

• Consequences of Drug Abuse for

Individual: Education, Employment, Income. Family : Violence. Society : Crime.

Nation : Law and Order problem.

UNIT-III

• Management of Drug Abuse

Medical Management: Medication for treatment of different types of drug abuses. Medication to reduce withdrawal effects.

UNIT-IV

• Psychiatric Management: Counseling, Behavioral and Cognitive therapy.

• Social Management: Family, Group therapy and Environmental Intervention.

References:

- 1. Ahuja, Ram (2003), Social Problems in India, Rawat Publication, Jaipur.
- 2. Extent, Pattern and Trend of Drug Use in India, Ministry of Social Justice and Empowerment, Government of India, 2004.
- 3. Inciardi, J.A. 1981. The Drug Crime Connection. Beverly Hills: Sage Publications. 23
- 4. Jasjit Kaur Randhawa & Samreet Randhawa, "Drug Abuse Problem, Management & Prevention", KLS, ISBN No. 978-81-936570-8-9, (2019).
- 5. Kapoor. T. (1985) Drug epidemic among Indian Youth, New Delhi: Mittal Pub.
- 6. Modi, Ishwar and Modi, Shalini (1997) Drugs: Addiction and Prevention, Jaipur: Rawat Publication.
- 7. Sain, Bhim 1991, Drug Addiction Alcoholism, Smoking obscenity New Delhi: Mittal Publications.
- 8. Sandhu, Ranvinder Singh, 2009, Drug Addiction in Punjab: A Sociological Study. Amritsar. Guru Nanak Dev University.
- 9. Singh, C. P. 2000. Alcohol and Dependence among Industrial Workers: Delhi: Shipra.
- 10. Sussman, S and Ames, S.L. (2008). Drug Abuse: Concepts, Prevention and Cessation, Cambridge University Press.
- 11. World Drug Report 2011, United Nations office of Drug and Crime.

Course Outcomes: The students will be able:

CO-1. To describe issues of cultural identity, ethnic background, age and gender in prevention, treatment and recovery.
CO-2. To describe warning sign, symptoms, and the course of substance use disorders.
CO-3. To describe principles and philosophy of prevention, treatment and recovery.
CO-4. To describe current and evidenced-based approaches practiced in the field of drug addiction.

SEMESTER–II BHP-121 ELECTRICITY & MAGNETISM-II

Time: 3 Hours

400 Maximum Marks: 100 (Theory Marks: 75+Internal Assessment: 25) Pass Marks: 35%

Credits: L T P

Note for paper setter and students:

- 1. There will be five sections.
- 2. Section A is compulsory and will be of 15 marks consisting of 8 short answer type questions carrying 2.5 mark each covering the whole syllabus. The answer should not exceed 50 words. The candidate will have to attempt any 6 questions in this section.
- **3.** Sections B, C, D and E will be set from units I, II, III & IV respectively and will consist of two questions of 15 marks each from the respective unit. The candidates are required to attempt one question from each of these sections. Each question in these sections should not have more than two subparts.
- 4. Non-Programmable Scientific calculator is allowed.

Course Objectives: Objectives of this course are to understand the magnetic effects of electric current. To understand magnetic properties of matter. To understand the concept of diamagnetism, paramagnetism and ferromagnetism. To understand the concept and applications of electromagnetic induction. To understand science and applications of electromagnetic waves.

Course Contents:

UNIT-I

Magnetic Effect of Electric Current; Direction of Field Lines due to current Flowing in a straight Conductor; Magnetic Field Density; Magnitude of Magnetic Flux; Magnetic and Lorentz Forces; Biot-Savart's Law; Magnetic Field Due to along Straight Conductor; Magnetic Field Intensity at point on the axis of a current loop; Variation of Field along the axis of the coil; Magnetic Field intensity inside a long Solenoid; Definition of Ampere; Ampere's Circuital Law; **Ampere's law and shape of path**; **Generalised form of Ampere's law**; Applications of Ampere's Circuital Law; **Comparison between Gauss's law and Ampere's law**; Solenoid Nature of Vector Field; Diversion of magnetic field; Properties of the Magnetic Field; **Magnetic scalar and Vector Potentials; expression for magnetic vector potential**; Surface Current Density; Changes in Magnetic Field at a current Sheet; Hall Effect,; Comparison of Electrostatic Field and Magnetic Field.

UNIT-II

Some Important Terms associated with Magnetic Materials; Torque on current Loop; Magnetic Dipole in a Magnetic Field; Potential Energy of Magnetic Dipole; Force on Magnetic Dipole in Non-

Uniform Magnetic Field; Magnetic Dipole Moment of an Atom; Expression of orbital Magnetic dipole moment of Electron; Electron Spin Magnetic Moment; Free and Bound Currents; Uniformly and Non Uniformly Magnetised **Materials**; Diamagnetism; Langevin's theory of diamagnetic behaviour; Paramagnetism and Langevin's Theory of Paramagnetic Susceptibility; Ferromagnetism; Domain theory of Magnetism; Hystersis Curve; Energy Loss Due to Hystersis and its Importance; **applications of hysteresis loop; current advances in magnetic materials.**

UNIT-III

Electromagnetic Induction; Faraday's Induction Experiments; Faraday's Laws of Electromagnetic Induction (Integral And Differential Forms); Lenz's law; **method of generating induced emf; mechanism of induced emf**; Self Induction; Neumann's Formula; Self Inductance of a Solenoid and a Torodial; Mutual Induction; Expression for Coefficient of Mutual Induction and Reciprocity theorem; Mutual Inductance of **two coils**, two **coaxial** Solenoids; Modification of Ampere's Law and the Displacement Current; Maxwell's Equation of Electromagnetism; Series and Parallel LCR Circuits; Average Power Associated With LCR Circuit

UNIT-IV

Mechanical waves versus electromagnetic waves; Production of em waves, EM wave spectrum, Linear, isotropic and homogenous dielectric medium; conducting and dielectric media; differential equations for electromagnetic waves; Nature of em waves; velocity of electromagnetic waves in vacuum; EM wave equation for a medium having finite μ and ε but $\sigma = 0$; Wave equation for plane polarized em waves and their solutions; Relation between electric and magnetic field of an em wave; electromagnetic wave propagation in a dielectric medium; Impedence of a dielectric to em waves, The Poynting vector and flow of energy in an em wave; Equation of continuity; EM waves for a medium having finite values of μ and ε but $\sigma \neq 0$; Solution of wave equation for a conducting medium; Skin depth; EM wave velocity and wave dispersion in a conductor; Behaviour of a medium as a conductor or dielectric; Characteristic impedance of a conducting medium to em waves; magnetic and electric energy densities; Poynting vector and Equation of Continuity for a Conducting medium; current advances in em waves.

TUTORIAL: Relevant problem on the topics covered in the course.

Books Prescribed

1. Electricity & Magnetism-T.S. Bhatia and Gurpreet Singh, Vishal Publishing Co.

- 2. Waves and Vibrations, T.S. Bhatia, Vishal Publishing Co.
- 3. Field & wave Electromagnetics by David & Cheng, Addison Wesley Publishing co

4. Electricity & Magnetism- A.K. Sikri, Pradeep Publications

Course	Outcomes:	
Cn No		

Sr. No.	On completing the course, the students will be able to:
CO1	Understand the magnetic effects of electric current.
CO2	Understand magnetic properties of matter.
CO3	Understand diamagnetism, paramagnetism and ferromagnetism.
CO4	Understand the concept and applications of electromagnetic induction.
CO5	Understand electromagnetic waves.

SEMESTER-II BHP-122 WAVES & OSCILLATIONS

Credits: L T P 400 Maximum Marks: 100 (Theory Marks: 75+Internal Assessment: 25) Pass Marks: 35%

Time: 3 Hours

Note for paper setter and students:

- 1. There will be five sections.
- 2. Section A is compulsory and will be of 15 marks consisting of 8 short answer type questions carrying 2.5 mark each covering the whole syllabus. The answer should not exceed 50 words. The candidate will have to attempt any 6 questions in this section.
- **3.** Sections B, C, D and E will be set from units I, II, III & IV respectively and will consist of two questions of 15 marks each from the respective unit. The candidates are required to attempt one question from each of these sections. Each question in these sections should not have more than two subparts.
- 4. Non-Programmable Scientific calculator is allowed.

Course Objectives: The purpose of the course is to understand the physical characteristics of SHM and obtaining solution of the oscillator using differential equations, to calculate logarithmic decrement relaxation time and quality factor of a harmonic oscillator. This course provides information to understand the difference between simple harmonic vibrations of same frequencies and different frequencies, wave equation and to understand the significance of transverse waves and longitudinal waves, coupled mechanical as well as electrical oscillators.

Course Contents:

UNIT-I

Simple and Damped Oscillations: Simple Harmonic Motion, Differential equation of SHM, Graphical representation of SHM, Energy of SHM, Compound pendulum, Torsional pendulum, Superposition of two perpendicular SHM, Lissajous figures–superposition of many SHM's, complex number notation and use of exponential series. Damped motion of mechanical and electrical oscillator, heavy damping, critical damping. Energy dissipation and energy of damped oscillator, amplitude decay, logarithmic decrement, relaxation time, Q-value, comparison between Free and Damped oscillations

UNIT-II

Forced Oscillations: Differential equation of forced mechanical oscillator, Transient and steady state behaviour of a forced oscillator, Variation of displacement and velocity with

frequency of driving force, frequency dependence of phase angle between force and displacement, velocity, Power supplied to oscillator by driving force and its variation with driving force frequency, Resonance absorption and Q-value as a measure of power absorption bandwidth, Q-value as amplification factor, Forced electrical oscillator, Variation of current with frequency, Variation of power supplied with frequency of applied voltage, Q-factor as amplification factor.

UNIT-III

Coupled Oscillations: Stiffness coupled oscillators, In phase and Out phase modes, normal coordinates and normal modes of vibration, solutions for differential equations for normal modes and exchange of energy, inductance coupling of electrical oscillators, loose, **intermediate and strong coupling**, energy exchange between two electrically coupled oscillators.

UNIT-IV

Wave Motion: Types of wave motion, wave equation, transverse waves on a string, the string as a forced oscillator, characteristic impedance of a string, reflection and transmission of transverse waves on a string at a boundary, Energy of a progressive wave, impedance matching, standing waves on a string of fixed length, **Energy of a vibrating string, normal modes and eigen frequencies,** Energy in a normal mode of oscillation wave groups, group velocity, dispersive and non-dispersive media, Longitudinal waves.

TUTORIAL: Relevant problem on the topics covered in the course.

Books Prescribed

1. Waves and Vibrations, T.S. Bhatia, Vishal Publishing Co.

2. The Physics of Vibrations and Waves- H.J. Pain, John Wiley and Sons Ltd , University of Chichester, 2005.

3. Waves and Oscillations – N. Subrahmanyam and Brij Lal, Vikas Publishing House Pvt. Ltd, 2013

Course Outcomes:

Sr. No.	On completing the course, the students will be able to:
CO1	Understand the physical characteristics of SHM and obtaining solution of the
	oscillator using differential equations.
CO2	Calculate logarithmic decrement relaxation time and quality factor of a harmonic oscillator.
CO3	Understand the difference between simple harmonic vibrations of same frequencies and different frequencies.
CO4	Solve wave equation and to understand the significance of transverse waves and longitudinal waves.
CO5	Explain the concept of coupled mechanical as well as electrical oscillators.

SEMESTER – II BHP-123 PHYSICS LAB-III

Time: 3 Hours

Credits: L T P 0 0 2 Maximum Marks: 50 (Theory Marks: 37+Internal Assessment: 13) Pass Marks: 35%

General Guidelines for Practical Examination

I. The distribution of marks is as follows:

i) One experiment **15 Marks**

ii) Brief Theory 5 Marks

iii) Viva-Voce 10Marks

iv) Record (Practical file) 7 Marks

II. There will be one sessions of 3 hours duration. The paper will have one session and will consist of 8 experiments out of which an examinee will mark 6 experiments and one of these is to be allotted by the external examiner.

III. Number of candidates in a group for practical examination should not exceed 12.

IV. In a single group no experiment be allotted to more than three examinee in any group.

Course Objectives: The purpose of the course is to understand the concept of resonance in series and parallel LCR circuits, study of induced e.m.f. as a function of velocity of a magnet, acceleration due to gravity by compound pendulum and by Kater's reversible pendulum. It includes measurement of logarithmic decrement, coefficient of damping, relaxation time and quality factor of a damped simple pendulum, laws of vibrating string by using Melde's apparatus and to show that $\lambda/2$ is constant.

LIST OF EXPERIMENTS

- 1. To study the magnetic field produced by a current carrying solenoid using a search coil and calculate permeability of air.
- 2. To study the induced e.m.f. as a function of the velocity of the magnet.
- 3. To study the phase relationships using impedance triangle for LCR circuit and calculate impedance.
- 4. Resonance in a series and parallel LCR circuits for different R-value and calculate Q- value.
- 5. To measure the charge sensitivity of a moving coil Ballistic galvanometer using a known capacitor.
- 6. To measure the self-inductance L of a given coil by Anderson Bridge method.
- 7. To find the value of B_H, the horizontal component of ear using a deflection & vibration magnetometer.
- 8. To study the variation of magnetic field with distance along the axis of coil carrying current by plotting a graph.

Books Prescribed

Practical Physics Vol. I, T.S. Bhatia, Gursharan Kaur, Iqbal Singh, Vishal Publications.
 Practical Physics, C.L. Arora, S. Chand & Co.

Course Outcomes:

Sr. No.	On completing the course, the students will be able to:
CO1	Understand the concept of resonance in series and parallel LCR circuits.
CO2	Study the induced e.m.f. as a function of velocity of a magnet.
CO3	Find out acceleration due to gravity by compound pendulum and by Kater's reversible pendulum.
CO4	Measure logarithmic decrement, coefficient of damping, relaxation time and quality factor of a damped simple pendulum.
CO5	Verify the laws of vibrating string by using Melde's apparatus and to show that $\lambda/2$ is constant.

SEMESTER-II BHP-124 PHYSICS LAB-IV

Credits: L T P

002

Time: 3 Hours

Maximum Marks: 50 (Theory Marks: 37+Internal Assessment: 13) Pass Marks: 35%

General Guidelines for Practical Examination

I. The distribution of marks is as follows:

v)One experiment 15 Marks

vi) Brief Theory 5 Marks

vii) Viva-Voce 10Marks

viii) Record (Practical file) **7 Marks**

II. There will be one sessions of 3 hours duration. The paper will have one session and will consist of 8 experiments out of which an examinee will mark 6 experiments and one of these is to be allotted by the external examiner.

III. Number of candidates in a group for practical examination should not exceed 12.

IV. In a single group no experiment be allotted to more than three examinee in any group.

Course Objectives: The purpose of the course is to understand the concept of resonance in series and parallel LCR circuits, study of induced e.m.f. as a function of velocity of a magnet, acceleration due to gravity by compound pendulum and by Kater's reversible pendulum. It includes measurement of logarithmic decrement, coefficient of damping, relaxation time and quality factor of a damped simple pendulum, laws of vibrating string by using Melde's apparatus and to show that $\lambda/2$ is constant.

LIST OF EXPERIMENTS

- 1. To plot a graph between the distance of the knife edge from the centre of gravity and the time period of a compound pendulum from graph find (a) acceleration due to gravity, (b) the radius of gyration and moment of inertia about an axis passing through centre of gravity.
- 2. To determine the acceleration due to gravity by Kater's reversible pendulum.
- 3. To verify the laws of vibrating string by using Meldes apparatus and to show that $\lambda/2$ is constant.
- 4. To compare the mass of two vibrating strings by using Meldes apparatus.
- 5. To measure logarithmic decrement, coefficient of damping, relaxation time and quality factor of a damped simple pendulum.
- 6. Study of the electrical vibrator.
- 7. Study the formation of waves on the computer in excel spreadsheet.
- 8. Study the superposition of waves on the computer in excel spreadsheet.

Books Prescribed

- 1. Practical Physics Vol. I, T.S. Bhatia, Gursharan Kaur, Iqbal Singh, Vishal Publications.
- 2. Practical Physics, C.L. Arora, S. Chand & Co.

Course Outcomes:

Sr. No.	On completing the course, the students will be able to:
CO1	Understand the concept of resonance in series and parallel LCR circuits.
CO2	Study the induced e.m.f. as a function of velocity of a magnet.
CO3	Find out acceleration due to gravity by compound pendulum and by Kater's reversible pendulum.
CO4	Measure logarithmic decrement, coefficient of damping, relaxation time and quality factor of a damped simple pendulum.
CO5	Verify the laws of vibrating string by using Melde's apparatus and to show that $\lambda/2$ is constant.

B.Sc. Physics (3 Years) / B.Sc. Physics (Honours) (4 Years)

SEMESTER-II MAP-121 MATHEMATICS -II

Time: 3 Hours

Credits: L T P 3 1 0 Maximum Marks: 100 (Max. Marks: 75+Internal Assessment: 25) Pass Marks: 35%

INSTRUCTIONS FOR PAPER SETTERS:

1. The question paper will consist of five sections namely Section-A which will be from entire syllabus (equally distributed from each unit) Section–B, C, D and E from Unit-I, II, III and IV, respectively.

2. The Section-A will consist of eight short answer type questions, each of 2.5 marks. Students are to attempt any six.

3. The Sections–B, C, D & E will consist of two questions each (each question should be subdivided into at most two parts). Students are to attempt any four questions in total by selecting one question from each section. Each question carries 15 marks.

4. Teaching time for this paper would be six periods per week.

COURSE OBJECTIVES:

- To acquire the knowledge of matrices.
- To have an idea about the two-dimensional geometry and its applications.
- To perceive knowledge about Polynomial equations.

Course Content:

UNIT-I

Coordinates Geometry(2D):Transformation of axes, shifting of origin, Rotation of axes, Parabola, Ellipse, Hyperbola and their properties; Tangent and normal, pair of tangents, Chord of contact for all the conics; Identifications of curves represented by second degree equation (without derivation)

UNIT-II

Solid Geometry: Straight line and planes in Intersection of two and three planes, Intersection of a line and plane; Sphere, Section of a sphere by a plane, Intersection of a line and sphere, Intersection of two spheres; Right circular Cone, Right circular Cylinder, Tangent lines, Tangent planes, and normal lines to these surfaces.

UNIT-III

Polynomial equations: Relation between the roots and co-efficients of polynomial equations (in one variable), Horner's method, Transformation of equations and symmetric functions of roots, Descartes rule of signs, Newton's method of divisors, Cardon's method, Solutions of biquadratic polynomial equations by Descartes and Ferrari's methods.

UNIT-IV

Introduction to Groups: Binary operations, Groups, Subgroups, Group table, Algebraic property, some standard algebraic properties (without proofs), Use of closure property, addition, identity and inverse with applications.

Reference Books:

1. Narayan, S.: Coordinate Geometry, Sultan Chand & Sons (2005).

- 2. Narayan, S.: Analytical Solid Geometry, Sultan Chand & Sons (2005).
- 3. Higher Engineering Mathematics: B.S. Grewal, Delhi, Khanna, 1995.
- 4. Mohan Singh, Topics in Maths, Lakshmi Publication, New Delhi, (1997)

5. N. S. Gopalakrishnan.: University Algebra, New Age International Publishers. (2007)

COURSE OUTCOMES: On completing the course students will be able to:

- 1. Understand the concept of axes.
- 2. Knowledge of parabola and its applications.
- 3. Implement knowledge of binary opeartions of groups.

CHX 121 **Inorganic Chemistry-II**

Credits: L T P

300

Maximum Marks: 75 (Theory Marks: 56+Internal Assessment: 19) Pass Marks: 35%

INSTRUCTIONS FOR PAPER SETTERS AND CANDIDATES (Scientific calculator is allowed)

- I. Examiner will make five sections of paper namely Section-I, II, III, IV and V
- Examiner will set total of NINE questions comprising ONE compulsory question of short answer type II. covering whole syllabi and TWO questions from each unit.
- Section-I will consist of eight six short questions carrying 2 marks each and students are required to III. attempt any six question.
- IV. Section-II, III, IV and V of paper will consist of EIGHT questions in total having TWO questions from each unit of the syllabus and each question carry 11 Marks.
- V. The students are required to attempt FIVE questions in all, taking ONE Compulsory question of section-I and one question from each section i.e. II, III, IV and V.

COURSE OBJECTIVES:

Time: 3 Hours

Students will learn naming of coordination complexes, Factors affecting co-ordination numbers and stereochemistry. The objective of the course is to teach the various theories dealing with the bonding in coordination compounds like VBT theory, CFT and MOT theory applied to homoneuclear diatomic molecules and heteronuclear Diatomic molecules.charge transfer transitions, π -Acid Ligands, and Alkali metal and alkaline earth metal chelators

COURSE CONTENTS:

UNIT-I Co-ordination Chemistry: Introduction, Werner's coordination theory, naming of co-ordinate complexes. Co-ordination numbers 1-12 and their stereo-chemistries. Factors affecting co-ordination numbers and stereo-chemistry

- **Configurational Isomers** (a)
- Conformational isomerism, (b)

Bonding in metal complexes: Valence bond theory for co-ordinate complexes, inner and outer orbital complexes, Electro-neutrality and back bonding, limitations of V.B. theory.

UNIT-II

Crystal field theory: Splitting of d-orbitals in octahedral, tetrahedral. Pairing Energy, Calculation of C.F.S.E. in high spin and low spin octahedral and High spin tetrahedral complexes, factors affecting the 10 Dq Value. Structural effects of crystal field splitting (Jahn-Teller distortion, variation of Ionic radii with increase in atomic number). Thermodynamics effects of C.F. splitting, variation in lattice energies and Hydration energies.

UNIT-III

Electronic spectra, Beer Lambert Law, Angular Momentum of electron spectra, Total angular momentum, Microstates and spectroscopic terms, a calculation of spectroscopic terms ford electronic configurations, L S coupling, Hund's rule

15Hrs

15Hrs

15Hrs

for finding the ground state terms, Electronic spectral properties of 1^{st} transition series, Orgel Diagrams for $d^1 - d^{10}$ systems, for weak field octahedral and tetrahedral complexes, limitations of C.F.T

UNIT-IV

15Hrs

Alkali metal and alkaline earth metal chelators: Macrocyclic ligands, macrocyclic effect, crownethers and podands, coronands, cryptands, structure of 18 crown-6 complex with KNCS, ion cavity complex, effect of anion and cation type on complex structure, simultaneous complexation of metal ion and water or of two metal ions, sandwich formation.

BOOKS PRESCRIBED:

- 1. J.E. Huheey, Inorganic Chemistry, 3rd Ed.
- 2. F.A. Cotton and G. Wilkinson, Advanced Inorganic Chemistry.
- 3. B.E. Douglas and D.H. McDaniel, Concepts and Models of Inorganic Chemistry.
- 4. R. Hilgenfeld and W. Saengar, Topics in current chemistry Vol-II.

COURSE OUTCOMES:

Sr. No.	On completing the course, the student will be able to
CO1	Students learned about the coordination compounds, theory, their nature of bonding,
CO2	Gain knowledge to apply ligand field theory CFT on simple molecules.
CO3	Learn about Molecular orbital theory
CO4	Learn about VSEPR theory, VBT
CO5	Learn and apply HSAB principle, Orgel Diagram, Macrocyclic ligands

SEMESTER – II CHP 122 Inorganic Chemistry Lab-II

Time: 3 Hours

Credits: L T P 0 0 1 Maximum Marks: 25 (Theory Marks: 19+Internal Assessment: 06) Pass Marks: 35%

INSTRUCTIONS FOR PAPER SETTERS AND CANDIDATES:

I. Examiner will give one organic salt to the students.

II. Each student will get different salt and analyse it for elements, functional group and prepare its derivatives.

III. The question paper will be 19 marks with split as under:(Write up = 6, Performance = 6, Viva-voce = 5, Practical note book = 2)

COURSE OBJECTIVE:

Students learn to identify and separate different cations in the inorganic mixtures through different methods. Students will be able to perform special tests for anions.

COURSE CONTENTS:

Section-A

Identification of cations and anions in a mixture which may contain combinations of acid ions.

a) Special Tests for Mixture of Anions

(i) Carbonate in the presence of sulphate.

(ii) Nitrate in the presence of nitrite

(iii) Nitrate in the presence of bromide and iodide.

(iv) Nitrate in the presence of chlorate.

(v) Chloride in the presence of bromide and iodide.

(vi) Chloride in the presence of iodide.

(vii) Bromide and iodide in the presence of each other and of chloride.

(viii) Phosphate, arsenate and arsenite in the presence of each other.

(ix) Sulphide, sulphite, thiosulphate and sulphate in the presence of each other.

(x) Borate in the presence of copper and barium salts.

(xi) Oxalate in the presence of fluoride.

Section-B

Identification of Cations in Mixtures

Identification of Group I, Group II (Group IIA and IIB), Group III, Group IV, Group V and Group VI cations.

BOOKS PRESCRIBED:

Vogel's book on Inorganic Qualitative Analysis

COURSE OUTCOMES:

Sr. No.	On completing the course, the student will be able to
CO1	Identify the anions present in the mixture.
CO2	Identify the cations present in the mixture.
CO3	Practice of handling different Chemicals in the lab
CO4	Learn to prepare basic solution required to identify cations and anions in the mixture
CO5	Learn about determination of boiling points of various compounds.

B.Sc. Physics (3 Years) / B.Sc. Physics (Honours) (4 Years) SEMESTER – II BCEN-1223 SEMESTER – II COMMUNICATIVE ENGLISH-II B.Sc.(Hons.) Physics, Chemistry, Botany, Zoology, Maths and Computational Statistics and Data Analytics

Credits: L T P

310

Time: 3 Hours Maximum Marks=100 (Theory 60 Marks +Practical 15 Marks+ Internal Assessment 25 Marks)

Pass Marks: 35%

Instructions for the Paper Setter and Distribution of Marks: The question paper will consist of four sections and the distribution of marks will be as under: Section A: 12 Marks Section B: 12 Marks Section C: 18 Marks Section D: 18 Marks Section A

1. Fifteen (15) Questions on the usage of Tenses, Conjunctions, and Subject-Verb Agreement will be set. The students will be required to attempt any Twelve (12).

(12X1=12 Marks)

Section **B**

2. ONE question (with sub parts) based on Skills and Strategies development exercises in Unit-3 and Unit-4 of the prescribed text book *Making Connections* will be set.

(1X12= 12marks)

Section C

3. Five short answer type questions from Unit 3 and 4 of *Making Connections : A Strategic Approach To Academic Reading* will be set. The students will be required to attempt any three. (3X2= 06 marks)

4. Four Essay type question (Two from each unit) from Unit 3 and 4 of *Making Connections: A strategic Approach to Academic Reading* will be set. The students will be required to answer any two, choosing at least one from each unit.

(2X6=12 marks)

Section D

5. Transcoding (given dialogue to prose or given prose to dialogue).

6. Taking notes on a speech/lecture/telephonic conversations.

(1X6=6 Marks)

(1X6 = 6 Marks)

7. Translation from Vernacular (Punjabi/ Hindi) to English (Isolated Sentences)

(1X6=6 Marks)

Course Objectives:

I: To develop competence in oral and visual communication.

II: To inculcate innovative and critical thinking among the students.

III: To enable them to grasp the application of communication theories.

IV: To acquire the knowledge of latest technology related with communication skills.

V: To provide knowledge of multifarious opportunities in the field of this programme.

Course Contents:

1. Reading and Comprehension Skills:

Students will be required to read and comprehend the essays in Unit 3 and 4 of the book *Making Connections: A Strategic Approach to Academic Reading* by Kenneth J. Pakenham, Third Edition.

2. Speaking and Conversational Skills: Components of a meaningful and easy conversation; understanding the cue and making appropriate responses; asking and providing information on general topics, situation based Conversation in English.

3. Grammar: Tenses, Conjunctions, and Subject-Verb Agreement.

Prescribed Books:

Making Connections by Kenneth J. Pakenham 3rd Edn. CUP

Recommended Books:

- 1. Oxford Guide to Effective Writing and Speaking by John Seely.
- 2. The Written Word by Vandana R Singh, Oxford University Press

Course Outcomes:

The completion of this course enables students to:

1. Identify common errors in language and rectify them.

2. Develop and expand writing skills through controlled and guided activities.

3. Develop coherence, cohesion and competence in oral discourse through intelligible pronunciation.

4. Develop the ability to handle the interview process confidently and learn the subtle nuances of an effective group discourse.

5. Communicate contextually in specific and professional situations with courtesy.

PRACTICAL (Marks: 15)

Course Contents:-

- 1. Oral Presentation. (5 Marks)
- 2. Group Discussion. (5 Marks)
- 3. Mock Interview (5 Marks)

SEMESTER – II BHPB-1201

Punjabi (Compulsory)-2

ਪੰਜਾਬੀ (ਲਾਜ਼ਮੀ)−2

Credit & Marks Distribution and Pre-Requisites of the Course

Course title & Code	Total Teaching Hours	Total Credits/ Hours per	Credit distribution			Total Marks 100		Time Allowed in Exam
		week	L	Т	Р	Theory	IA	
ਪੰਜਾਬੀ (ਲਾਜ਼ਮੀ)−2 BHPB-1201	60	4	4	0	0	75	25	3 Hours

ਕੋਰਸ ਦਾ ਉਦੇਸ਼ Course Objective	ਪਾਠ-ਕ੍ਰਮ ਨਤੀਜੇ Course Outcomes (COs)
 ਵਿਦਿਆਰਥੀਆਂ ਵਿਚ ਸਾਹਿਤਕ ਰੁਚੀਆਂ ਪੈਦਾ ਕਰਨਾ। 	 ਉਸ ਅੰਦਰ ਸਾਹਿਤਕ ਰੁਚੀਆਂ ਪ੍ਰਫੁਲਿੱਤ ਹੋਣਗੀਆਂ।
 ਆਲੋਚਨਾਤਮਕ ਰੁਚੀਆਂ ਨੂੰ ਵਿਕਸਤ ਕਰਨਾ। 	 ਉਸ ਅੰਦਰ ਸਾਹਿਤ ਸਿਰਜਣਾ ਦੀ ਸੰਭਾਵਨਾ ਵਧੇਗੀ।
 ਵਿਦਿਆਰਥੀ ਨੂੰ ਦਫ਼ਤਰੀ ਅਤੇ ਘਰੇਲੂ ਚਿੱਠੀ ਪੱਤਰ ਤੋਂ ਜਾਣੂ 	 ਵਿਦਿਆਰਥੀ ਚਿੱਠੀ-ਪੱਤਰ ਦੀ ਲਿਖਣ ਸ਼ੈਲੀ ਤੋਂ ਜਾਣ ਹੋਵੇਗਾ।
ਕਰਵਾਉਣਾ।	 ਉਹ ਭਾਸ਼ਾਈ ਬਣਤਰ ਤੋਂ ਜਾਣ ਹੋਵੇਗਾ।
 ਭਾਸ਼ਾਈ ਗਿਆਨ ਵਿਚ ਵਾਧਾ ਕਰਨਾ। 	

ਅੰਕ–ਵੰਡ ਅਤੇ ਪ੍ਰੀਖਿਅਕ ਲਈ ਹਦਾਇਤਾਂ

ਸਿਲੇਬਸ ਦੇ ਚਾਰ ਭਾਗ ਹਨ ਪਰ ਪ੍ਰਸ਼ਨ-ਪੱਤਰ ਦੇ ਪੰਜ ਭਾਗ ਹੋਣਗੇ। ਪਹਿਲੇ ਭਾਗ ਵਿਚ 1.5-1.5 (ਡੇਢ-ਡੇਢ) ਅੰਕ ਦੇ ਅਤਿ-ਸੰਖੇਪ (Objective Type) 10 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ ਜੋ ਕਿ ਸਾਰੇ ਸਿਲੇਬਸ ਵਿਚੋਂ ਹੋਣਗੇ ਅਤੇ ਸਾਰੇ ਪ੍ਰਸ਼ਨ ਹੱਲ ਕਰਨੇ ਲਾਜ਼ਮੀ ਹੋਣਗੇ। ਸਿਲੇਬਸ ਦੇ ਬਾਕੀ ਚਾਰ ਭਾਗਾਂ ਵਿਚ 02-02 ਲੇਖ ਨੁਮਾ ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ। ਹਰੇਕ ਭਾਗ ਵਿਚੋਂ 01-01 ਪ੍ਰਸ਼ਨ ਕਰਨਾ ਲਾਜ਼ਮੀ ਹੋਵੇਗਾ। ਹਰੇਕ ਪ੍ਰਸ਼ਨ ਦੇ ਬਰਾਬਰ 15 ਅੰਕ ਹੋਣਗੇ। ਪੇਪਰ ਸੈੱਟਰ ਜੇਕਰ ਚਾਹੇ ਤਾਂ ਪ੍ਰਸ਼ਨਾਂ ਦੀ ਵੰਡ ਅੱਗੋਂ ਵੱਧ ਤੋਂ ਵੱਧ ਚਾਰ ਉਪ-ਪ੍ਰਸ਼ਨਾਂ ਵਿਚ ਕਰ ਸਕਦਾ ਹੈ।

ਪਾਠ–ਕ੍ਰਮ ਭਾਗ–ਪਹਿਲਾ

ਵਾਰਤਕ ਦੇ ਰੰਗ, (ਨਿਬੰਧ ਅਤੇ ਰੇਖਾ-ਚਿਤਰ) (ਸੰਪਾਦਕ) ਡਾ. ਮਹਿਲ ਸਿੰਘ, ਕਸਤੂਰੀ ਲਾਲ ਐਂਡ ਸਨਜ਼, ਅੰਮ੍ਰਿਤਸਰ। (ਨਿਬੰਧ ਭਾਗ ਵਿਚੋਂ ਸਾਰ/ਵਿਸ਼ਾ-ਵਸਤੁ। ਰੇਖਾ-ਚਿਤਰ ਭਾਗ ਵਿਚੋਂ ਸਾਰ/ਨਾਇਕ ਬਿੰਬ)

ਭਾਗ–ਦੂਜਾ

ਪੰਜਾਬ ਦੇ ਮਹਾਨ ਕਲਾਕਾਰ (ਬਲਵੰਤ ਗਾਰਗੀ) ਗੁਰੂ ਨਾਨਕ ਦੇਵ ਯੂਨੀਵਰਸਿਟੀ, ਅੰਮ੍ਰਿਤਸਰ। (ਸਤੀਸ਼ ਗੁਜਰਾਲ ਤੋਂ ਸੁਰਿੰਦਰ ਕੌਰ ਤਕ) (ਵਿਸ਼ਾ-ਵਸਤੁ/ਸਾਰ/ਨਾਇਕ ਬਿੰਬ)

ਭਾਗ–ਤੀਜਾ

(ੳ) ਦਫ਼ਤਰੀ ਚਿੱਠੀ ਪੱਤਰ

(ਅ) ਮੁਹਾਵਰੇ ਅਤੇ ਅਖਾਣ

ਭਾਗ–ਚੌਥਾ

(ੳ) ਸ਼ਬਦ-ਬਣਤਰ ਅਤੇ ਸ਼ਬਦ-ਰਚਨਾ - ਪਰਿਭਾਸ਼ਾ ਅਤੇ ਮੁਢਲੇ ਸੰਕਲਪ
(ਅ) ਸ਼ਬਦ-ਸ਼੍ਰੇਣੀਆਂ

SEMESTER – II BPBI-1202 Basic Punjabi -2 ਮੁਢਲੀ ਪੰਜਾਬੀ-2

(In Lieu of Compulsory Punjabi)

Credit & Marks Distribution and Pre-Requisites of the Course

Course title & Code	TotalTotalTeachingCredits /HoursHours perweekK	Credit distribution		Total Marks 100		Time Allowed in Exam		
		Hours per week	L	Т	Р	Theory	IA	
ਮੁਢਲੀ ਪੰਜਾਬੀ−2 BPBI-1202	60	4	4	0	0	75	25	3 Hours

E				
l	ਕੋਰਸ ਦਾ ਉਦੇਸ਼ Course Objective	ਪਾਠ–ਕ੍ਰਮ ਨਤੀਜੇ Course Outcomes (COs)		
	 ਵਿਦਿਆਰਥੀ ਅੰਦਰ ਸ਼ਬਦ ਬਣਤਰ ਦੀ ਸਮਝ ਵਿਕਸਤ ਕਰਨਾ। 	• ਉਹ ਪੰਜਾਬੀ ਸ਼ਬਦ-ਬਣਤਰ ਦੀ ਜਾਣਕਾਰੀ ਹਾਸਲ ਕਰਕੇ ਭਾਸ਼ਾਈ		
	 ਵਿਦਿਆਰਥੀ ਨੂੰ ਸ਼ਬਦ ਪ੍ਰਕਾਰ ਬਾਰੇ ਜਾਣਕਾਰੀ ਪ੍ਰਦਾਨ ਕਰਨਾ। 	ਗਿਆਨ ਨੂੰ ਵਿਕਸਿਤ ਕਰਨਗੇ।		
	 ਪੰਜਾਬੀ ਭਾਸ਼ਾ ਦੇ ਵਿਆਕਰਨਕ ਪ੍ਰਬੰਧ ਸਬੰਧੀ ਗਿਆਨ ਕਰਾਉਣਾ। 	 ਪੰਜਾਬੀ ਸ਼ਬਦ-ਰਚਨਾ ਸਬੰਧੀ ਮੁਹਾਰਤ ਹਾਸਲ ਕਰਨਗੇ। 		
	 ਸਿਖਲਾਈ ਤੇ ਅਭਿਆਸ ਦੁਆਰਾ ਪੰਜਾਬੀ ਸ਼ਬਦ ਭੰਡਾਰ 	 ਵਿਦਿਆਰਥੀ ਸ਼ਬਦਾਂ ਦੀਆਂ ਭਿੰਨ-ਭਿੰਨ ਕਿਸਮਾਂ ਤੋਂ ਜਾਣੂ ਹੋਵੇਗਾ। 		
	ਵਧਾਉਣਾ।	• ਵਿਦਿਆਰਥੀਆਂ 'ਚ ਨਿੱਤ ਵਰਤੋਂ ਦੀ ਪੰਜਾਬੀ ਸ਼ਬਦਾਵਲੀ ਭੰਡਾਰ 'ਚ		
		ਵਾਧਾ ਹੋਵੇਗਾ।		

ਅੰਕ–ਵੰਡ ਅਤੇ ਪ੍ਰੀਖਿਅਕ ਲਈ ਹਦਾਇਤਾਂ

ਸਿਲੇਬਸ ਦੇ ਚਾਰ ਭਾਗ ਹਨ ਪਰ ਪ੍ਰਸ਼ਨ-ਪੱਤਰ ਦੇ ਪੰਜ ਭਾਗ ਹੋਣਗੇ। ਪਹਿਲੇ ਭਾਗ ਵਿਚ 01-01 ਅੰਕ ਦੇ ਅਤਿ-ਸੰਖੇਪ ਉੱਤਰ ਵਾਲੇ (Objective Type) 11 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ ਜੋ ਕਿ ਸਾਰੇ ਸਿਲੇਬਸ ਵਿਚੋਂ ਹੋਣਗੇ ਅਤੇ ਸਾਰੇ ਪ੍ਰਸ਼ਨ ਹੱਲ ਕਰਨੇ ਲਾਜ਼ਮੀ ਹੋਣਗੇ।ਪ੍ਰਸ਼ਨ ਪੱਤਰ ਦੇ ਦੂਸਰੇ ਅਤੇ ਤੀਸਰੇ ਭਾਗ ਵਿਚ, ਸਿਲੇਬਸ ਦੇ ਪਹਿਲੇ ਅਤੇ ਦੂਸਰੇ ਭਾਗ ਵਿਚੋਂ 8-8 ਅੰਕਾਂ ਦੇ 3-3 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ। ਜਿੰਨ੍ਹਾਂ ਵਿਚੋਂ ਵਿਦਿਆਰਥੀ ਨੇ ਕੋਈ 2-2 ਪ੍ਰਸ਼ਨ ਹੱਲ ਕਰਨੇ ਹੋਣਗੇ। ਇਸੇ ਤਰ੍ਹਾਂ ਪ੍ਰਸ਼ਨ ਪੱਤਰ ਦੇ ਚੌਥੇ ਭਾਗ ਵਿਚ 4-4 ਅੰਕਾਂ ਦੇ 5 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ। ਜਿੰਨ੍ਹਾਂ ਵਿਚੋਂ ਵਿਦਿਆਰਥੀ ਨੇ 4 ਪ੍ਰਸ਼ਨ ਹੱਲ ਕਰਨੇ ਹੋਣਗੇ।ਭਾਗ ਪੰਜਵੇਂ ਵਿਚ 8-8 ਅੰਕਾਂ ਦੇ 3 ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ।ਜਿੰਨ੍ਹਾਂ ਵਿਚੋਂ ਵਿਦਿਆਰਥੀ ਨੇ 2 ਪ੍ਰਸ਼ਨ ਕਰਨੇ ਲਾਜ਼ਮੀ ਹੋਣਗੇ।

ਪਾਠ–ਕ੍ਰਮ ਭਾਗ–ਪਹਿਲਾ

ਪੰਜਾਬੀ ਸ਼ਬਦ-ਬਣਤਰ:

ਧਾਤੁ, ਵਧੇਤਰ (ਅਗੇਤਰ, ਮਧੇਤਰ, ਪਿਛੇਤਰ), ਪੰਜਾਬੀ ਕੋਸ਼ਗਤ ਸ਼ਬਦ ਅਤੇ ਵਿਆਕਰਨਕ ਸ਼ਬਦ

ਭਾਗ–ਦੂਜਾ

ਪੰਜਾਬੀ ਸ਼ਬਦ-ਪ੍ਰਕਾਰ:

(ੳ) ਸੰਯੁਕਤ ਸ਼ਬਦ, ਸਮਾਸੀ ਸ਼ਬਦ, ਦੋਜਾਤੀ ਸ਼ਬਦ, ਦੋਹਰੇ/ਦੁਹਰੁਕਤੀ ਸ਼ਬਦ ਅਤੇ ਮਿਸ਼ਰਤ ਸ਼ਬਦ

(ਅ) ਸਿਖਲਾਈ ਤੇ ਅਭਿਆਸ

ਭਾਗ–ਤੀਜਾ

ਪੰਜਾਬੀ ਸ਼ਬਦ-ਰਚਨਾ:

ਇਕ-ਵਚਨ/ਬਹੁ-ਵਚਨ, ਲਿੰਗ-ਪੁਲਿੰਗ, ਬਹੁਅਰਥਕ ਸ਼ਬਦ, ਸਮਾਨਅਰਥਕ ਸ਼ਬਦ, ਬਹੁਤੇ ਸ਼ਬਦਾਂ ਲਈ ਇਕ ਸ਼ਬਦ, ਸ਼ਬਦ ਜੁੱਟ, ਵਿਰੋਧਅਰਥਕ ਸ਼ਬਦ, ਸਮਨਾਮੀ ਸ਼ਬਦ

ਭਾਗ–ਚੌਥਾ

ਨਿੱਤ ਵਰਤੋਂ ਦੀ ਪੰਜਾਬੀ ਸ਼ਬਦਾਵਲੀ

ਖਾਣ-ਪੀਣ, ਸਾਕਾਦਾਰੀ, ਰੁੱਤਾਂ, ਮਹੀਨਿਆਂ, ਗਿਣਤੀ, ਮੌਸਮ, ਬਜ਼ਾਰ, ਵਪਾਰ, ਧੰਦਿਆਂ ਨਾਲ ਸੰਬੰਧਿਤ

SEMESTER – II BPHC-1204 PUNJAB HISTORY & CULTURE (C 321 TO 1000 A.D.) (Special Paper in lieu of Punjabi compulsory) (For those students who are not domicile of Punjab)

B. A.; B.A. (SS); B. A. (Hons. – English); B. Com. (Hons., R, Ac. & Finance); B. Sc. Bio-Tech./Comp. Sc./Eco./FD/Food Sc./IT/Med./N.Med.; B.Sc. (Hons. –Botany, Chemistry, Mathematics, Physics, Zoology); B. of Mult.; B. in Int. & Mob. Tech.; BBA; BCA; BJMC; B. Voc. (Software Development, Theatre and Stage Craft, Food Processing, Textile Design & Apparel Technology) BTTM (Bachelor of Tour and Travel Management)

Credits: L T P 400 Maximum Marks: 100 (Theory Marks: 75+Internal Assessment: 25) Pass Marks: 35%

Time: 3 Hours

Instructions for the Paper Setters:

Question paper should consist of two sections—Section A and Section B. The paper setter must ensure that questions in Section—A do not cover more than one point, and questions in Section—B should cover at least 50 per cent of the theme.

Section–A: The examiner will set 15 objective type questions out of which the candidate shall attempt any 10 questions, each carrying $1\frac{1}{2}$ marks. The total weightage of this section will be 15 marks. Answer to each question should be in approximately one to two sentences.

Section–B: The examiner will set 8 questions, two from each Unit. The candidate will attempt 4 questions selecting one from each Unit in about 1000 words. Each question will carry 15 marks. The total weightage of this section will be 60 marks.

Note: The examiner is to set the question paper in two languages: English & Hindi.

Course Objectives: The main objective of this course is to educate the students who are not domicile of the Punjab about the history and culture of the Ancient Punjab. It is to provide them knowledge about the social, economic, religious, cultural and political life of the peopleof the Punjab during the rule of various dynasties such as The Mauryans, The Khushans, The Guptas, The Vardhanas and other ancient ruling dynasties of the period under study.

Unit-I

- 1. The Punjab under Chandragupta Maurya and Ashoka.
- 2. The Kushans and their Contribution to the Punjab.

Unit-II

3. The Punjab under the Gupta Emperors.

4. The Punjab under the Vardhana Emperors

Unit-III

- 5. Political Developments 7th Century to 1000 A.D.
- 6. Socio-cultural History of Punjab from 7th Century to 1000 A.D.

Unit-IV

- 7. Development of languages and Literature.
- 8. Development of art & Architecture.

Suggested Readings:-

L. Joshi (ed.), *History and Culture of the Punjab*, Part-I, Patiala, 1989 (3rdedition).

L.M. Joshi and Fauja Singh (ed), *History of Punjab*, Vol.I, Patiala 1977.Budha Parkash, *Glimpses of Ancient Punjab*, Patiala, 1983.B.N. Sharma, *Life in Northern India*, Delhi. 1966.

Course Outcomes:

On completing the course, the students will be able to:

- **CO-1** Understand the history and culture of the Punjab in Ancient Period.
- **CO-2** Analyse social, economic, religious, cultural and political life of Ancient Indian dynasties.
- **CO-3** Study about the political developments from 7th century to 1000AD.
- **CO-4** Understand socio-cultural history of the Punjab from 7th century to 1000 AD.
- **CO-5** Analyse language, literature, art and architecture of Ancient Punjab.

SEMESTER-II Course Code: ZDA121 Course Title-DRUG ABUSE: PROBLEM, MANAGEMENT AND PREVENTION-II (Compulsory for all Under Graduate Classes)

Credits: L T P 100 Maximum Marks: 25

Time: 3 Hours

Instructions for the Paper Setters:

- 1) There will be two sections A and B.
- 2) Section A is compulsory and will be of 5 marks consisting of 8 short answer type questions carrying 1 marks each covering the whole syllabus. The candidates are required to attempt 5 questions out of 8 short answer type questions. The answer should not exceed 50 words.
- 3) Candidates shall be required to attempt 4 questions from Section B, selecting one question from each unit and each question carries 5 marks. Preferably, the question should not be split into more than two sub-parts.

Course Objectives:

The course aim is to

CO-1.	Describe the role of family in the prevention of drug abuse.
CO-2.	Describe the role of school and teachers in the prevention of drug abuse.
CO-3.	Emphasize the role of media and educational and awareness program.
CO-4.	Provide knowhow about various legislation and Acts against drug abuse.

UNIT-I

Role of family: Parent child relationship, Family support, Supervision, Shaping values, Active Scrutiny.

UNIT-II

School: Counselling, Teacher as role-model.

Parent-Teacher-Health Professional Coordination, Random testing on students.

UNIT-III

Controlling Drug Abuse: Media: Restraint on advertisements of drugs, advertisements on bad effects of drugs, Publicity and media, Campaigns against drug abuse, Educational and awareness program

UNIT-IV

Legislation: NDPS act, Statutory warnings, Policing of Borders, Checking Supply/Smuggling of Drugs, Strict enforcement of laws, Time bound trials.

References:

- 1. Ahuja, Ram (2003), Social Problems in India, Rawat Publication, Jaipur.
- 2. Extent, Pattern and Trend of Drug Use in India, Ministry of Social Justice and Empowerment, Government of India, 2004.
- 3. Inciardi, J.A. 1981. The Drug Crime Connection. Beverly Hills: Sage Publications.

- 4. Jasjit Kaur Randhawa & Samreet Randhawa, "Drug Abuse Problem, Management & Prevention", KLS, ISBN No. 978-81-936570-8-9, (2019).
- 5. Kapoor. T. (1985) Drug epidemic among Indian Youth, New Delhi: Mittal Pub.
- 6. Modi, Ishwar and Modi, Shalini (1997) Drugs: Addiction and Prevention, Jaipur: Rawat Publication.
- 7. Sain, Bhim 1991, Drug Addiction Alcoholism, Smoking obscenity New Delhi: Mittal Publications.
- 8. Sandhu, Ranvinder Singh, 2009, Drug Addiction in Punjab: A Sociological Study. Amritsar. Guru Nanak Dev University.
- 9. Singh, C. P. 2000. Alcohol and Dependence among Industrial Workers: Delhi: Shipra.
- 10. Sussman, S and Ames, S.L. (2008). Drug Abuse: Concepts, Prevention and Cessation, Cambridge University Press.
- 11. World Drug Report 2011, United Nations office of Drug and Crime.

Course Outcomes: The students will be able to:

CO-1.	Understand the importance of family and its role in drug abuse prevention.
CO-2.	Understand the role of support system especially in schools and inter-relationships
	between students, parents and teachers.
CO-3.	Understand impact of media on substance abuse prevention.
CO-4.	Understand the role of awareness drives, campaigns etc. in drug abuse management.
CO-5	Learn about the Legislations and Acts governing drug trafficking and Abuse in India.